

Strain-Promoted Reactions of 1,2,3-Cyclohexatriene and Its Derivatives

Nature **2023**, 618, 748-754

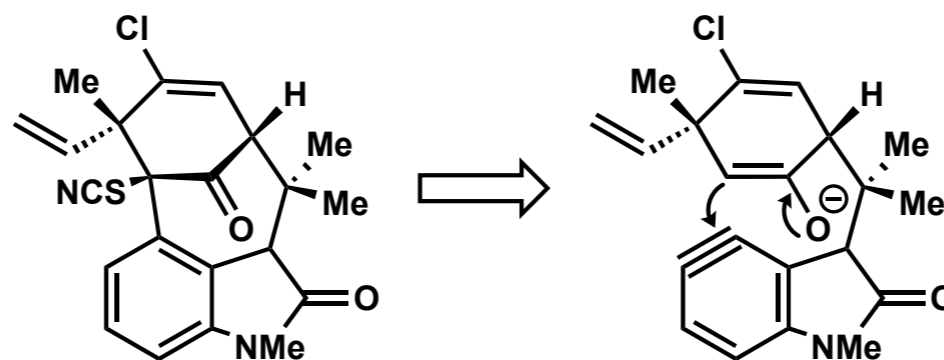
Samuel Tandang Kasmali G1

A Career in Strained Intermediates

2

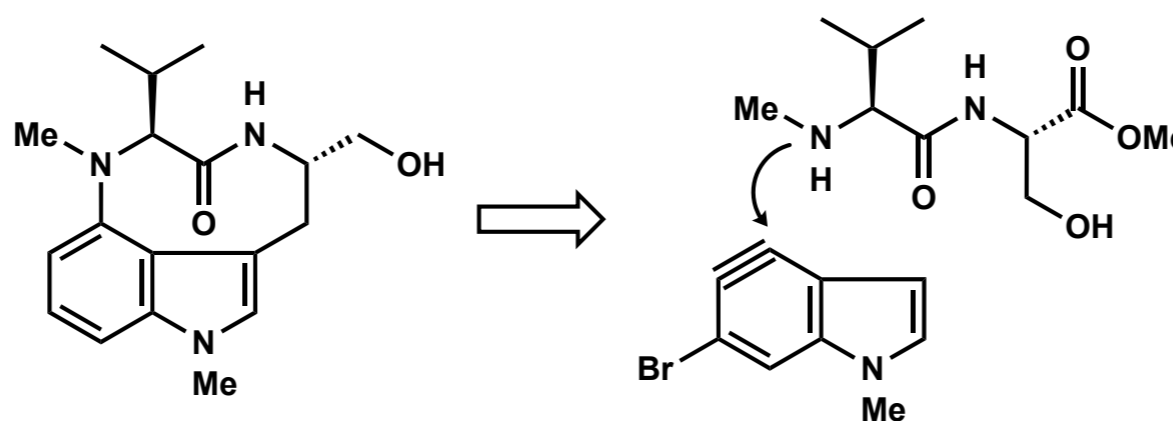


GARG LAB



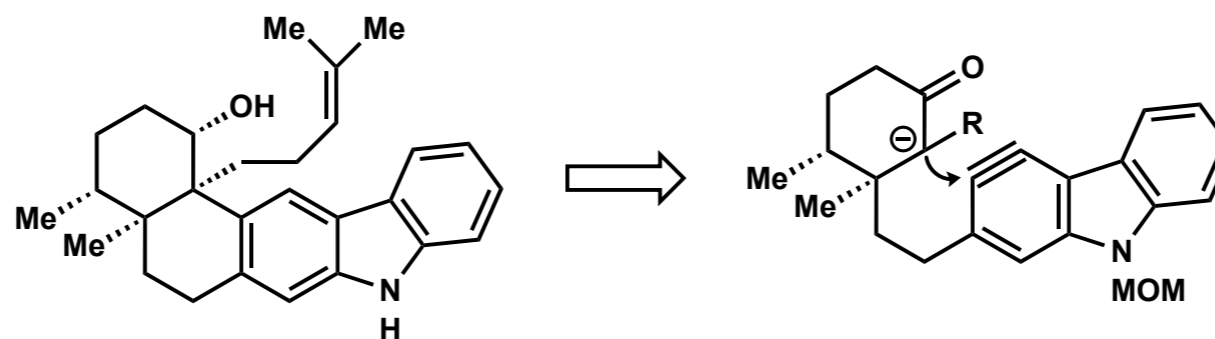
N-Methylwelwitindolinone C
Isothiocyanate

OL **2009**, 11, 2349–2351
JACS **2011**, 133, 15797–15799
JACS **2012**, 134, 1396–1399
ACIE **2012**, 51, 3758
ACIE **2013**, 52, 12422–12425
JACS **2014**, 136, 14710–14713



Indolactam V

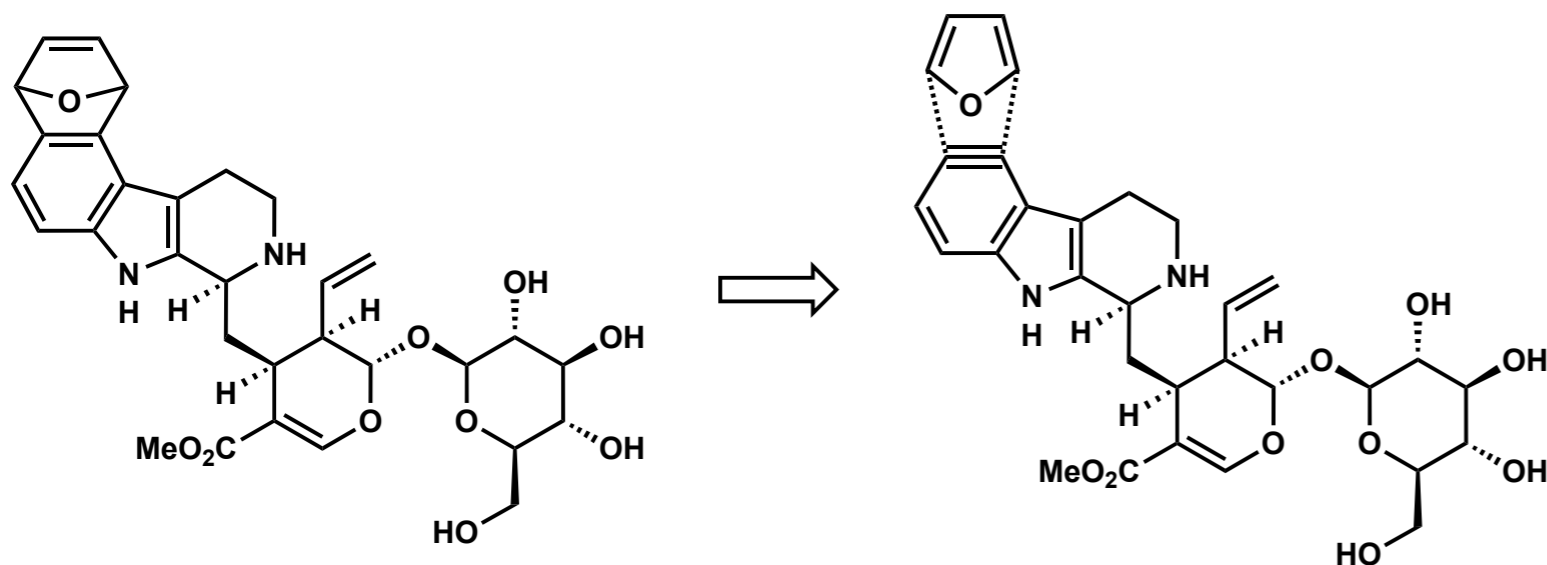
JACS **2011**, 133, 3832–3835
Synlett **2011**, 18, 2599–2604



Tubingensin A

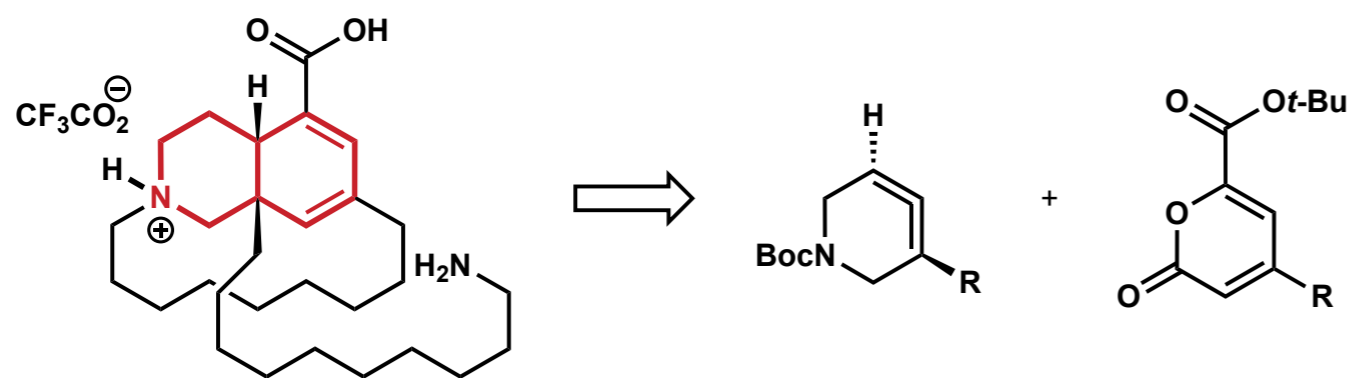
JACS **2014**, 136, 3036–3039

A Career in Strained Intermediates



JACS **2021**, *143*, 7471–7479

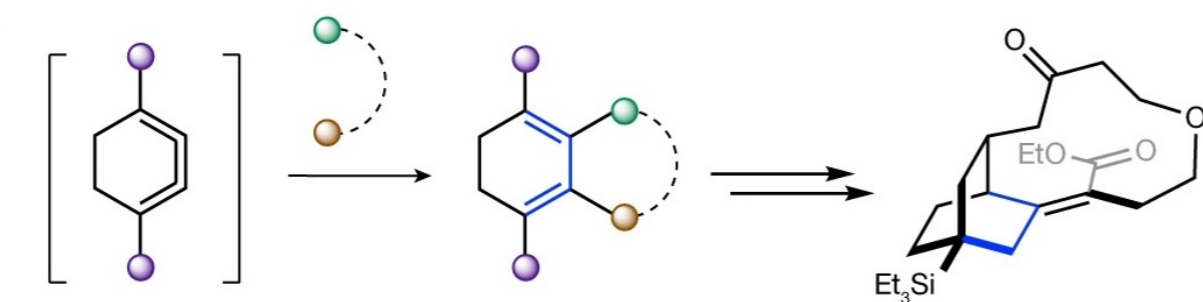
Strictosidine “*strictosidyne*”



Science **2023**, *379*, 261–265

Lissodendoric Acid A

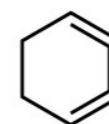
[This work]



From readily accessible precursors under mild conditions

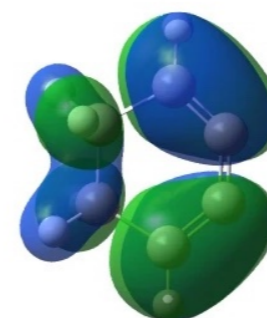
Diverse highly functionalized 1,3-dienes

Unconventional entryway to complex scaffolds

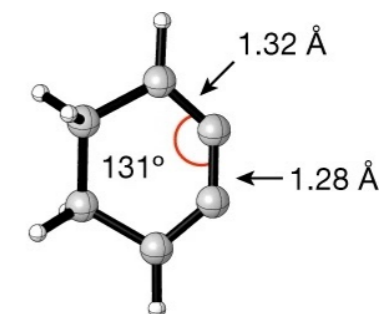


5

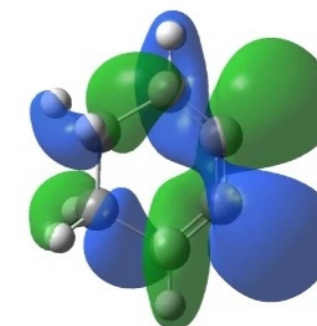
HOMO



(−6.2 eV)

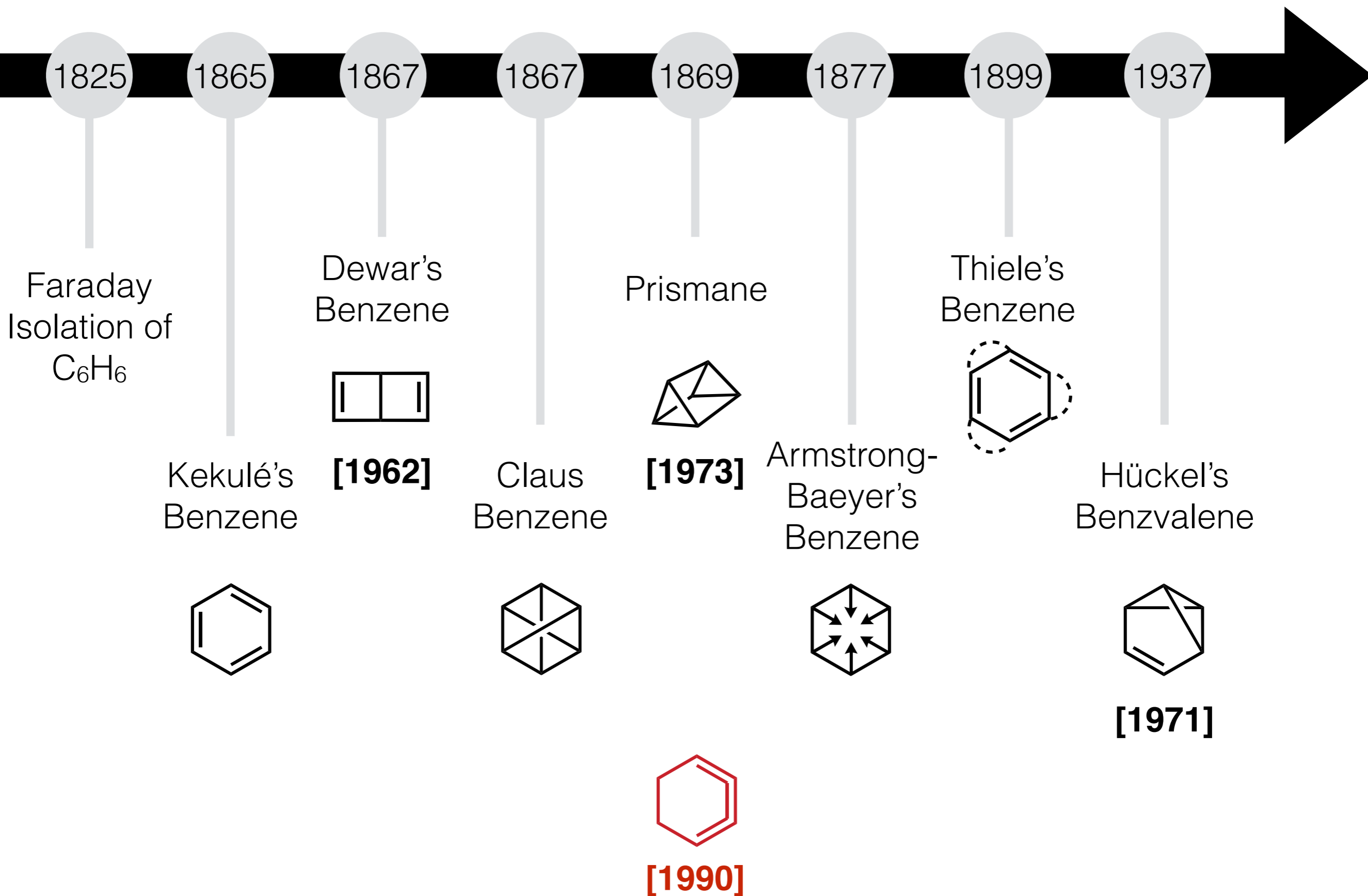


LUMO

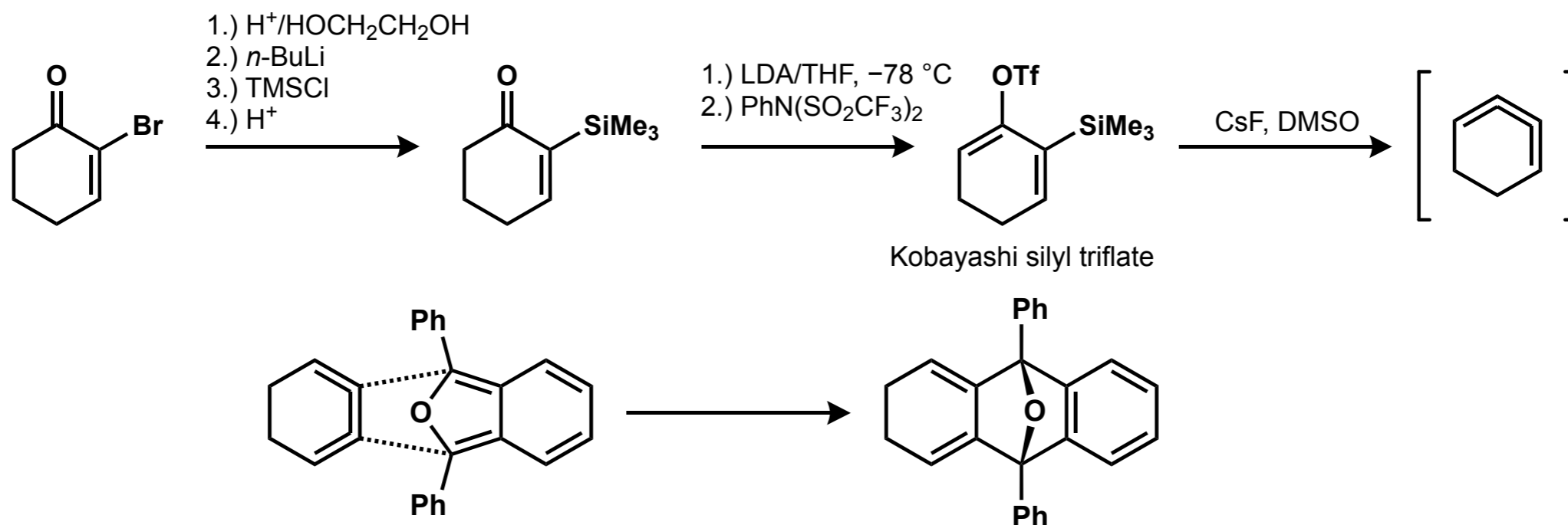


(−2.1 eV)

The Story of C₆H₆



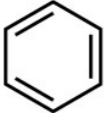
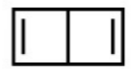



1,2,3-Cyclohexatriene Seminal Methods

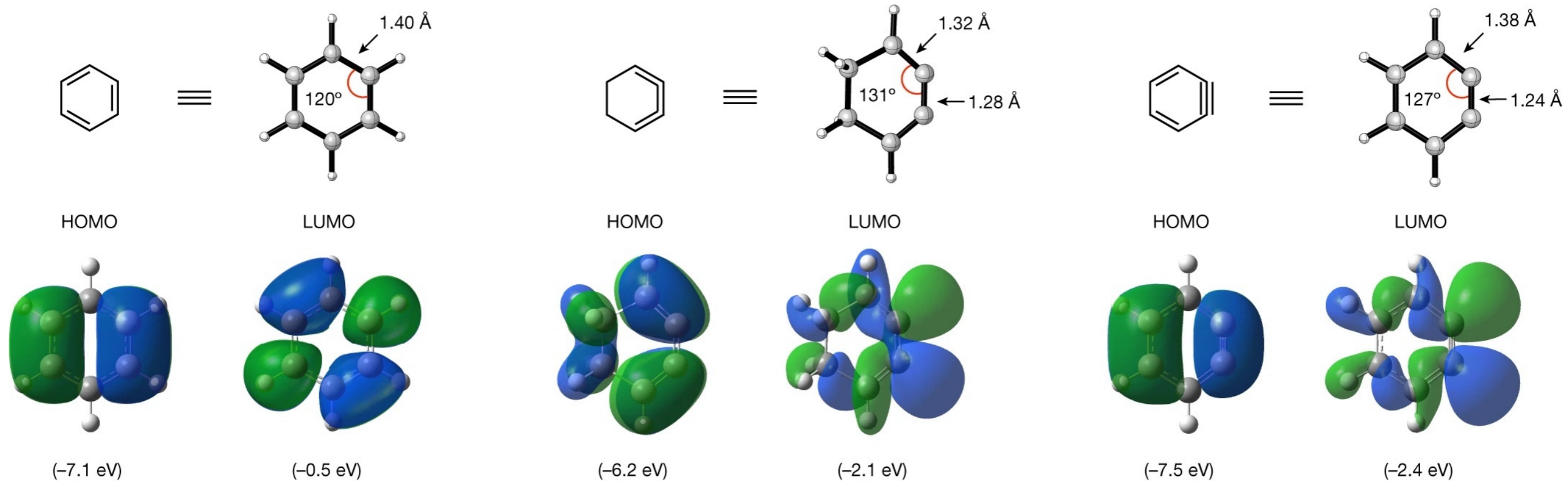


The most distinct advantage of this generation method is not because it CAN generate aryne but the way how it maintains a low-level concentration of aryne species under arynophile-friendly conditions. The combination of CsF and acetonitrile solvent, sometimes with toluene as cosolvent, is a magic recipe in Kobayashi's protocol. An explanation for this is that CsF has low solubility in acetonitrile, which in turn would only activate a small portion of *o*-silylaryl triflate at any reaction stage.⁶⁹ In addition, DFT calculations on the fluoride-induced benzyne generation process of *o*-silylphenyl triflate revealed that the removal of the TMS group by fluoride is through a pseudo-S_N2 mechanism with the formation of a pentacoordinated silicon ate complex as the rate-determining step.⁷⁰ Therefore, this system could provide a constant supply of a suitably low concentration of aryne intermediate. As a consequence of Kobayashi's method, the reaction efficiency can be generally enhanced with high functional group tolerance, which could also accommodate different aryne reaction modes.

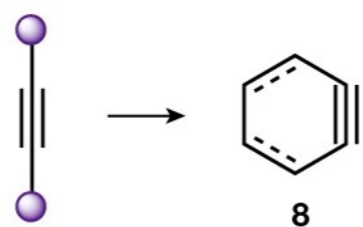
Computational Studies

LOT B3LYP-D3/6-311+G(d,p)

				
1 Benzene	2 Dewar benzene	3 Benzvalene	4 Prismane	5 1,2,3-Cyclohexatriene
(1825)	(1963)	(1966)	(1973)	(1990)
$\Delta G_{rel} = 0 \text{ kcal mol}^{-1}$	83 kcal mol ⁻¹	81 kcal mol ⁻¹	123 kcal mol ⁻¹	101 kcal mol ⁻¹

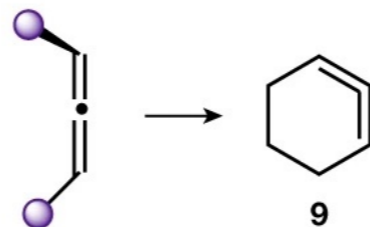


Arynes and cyclic alkynes



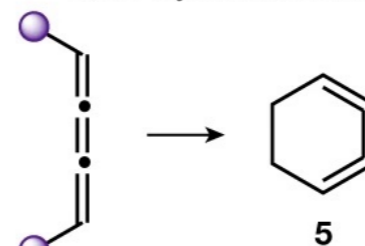
40–50 kcal mol⁻¹

Cyclic allenes



30 kcal mol⁻¹

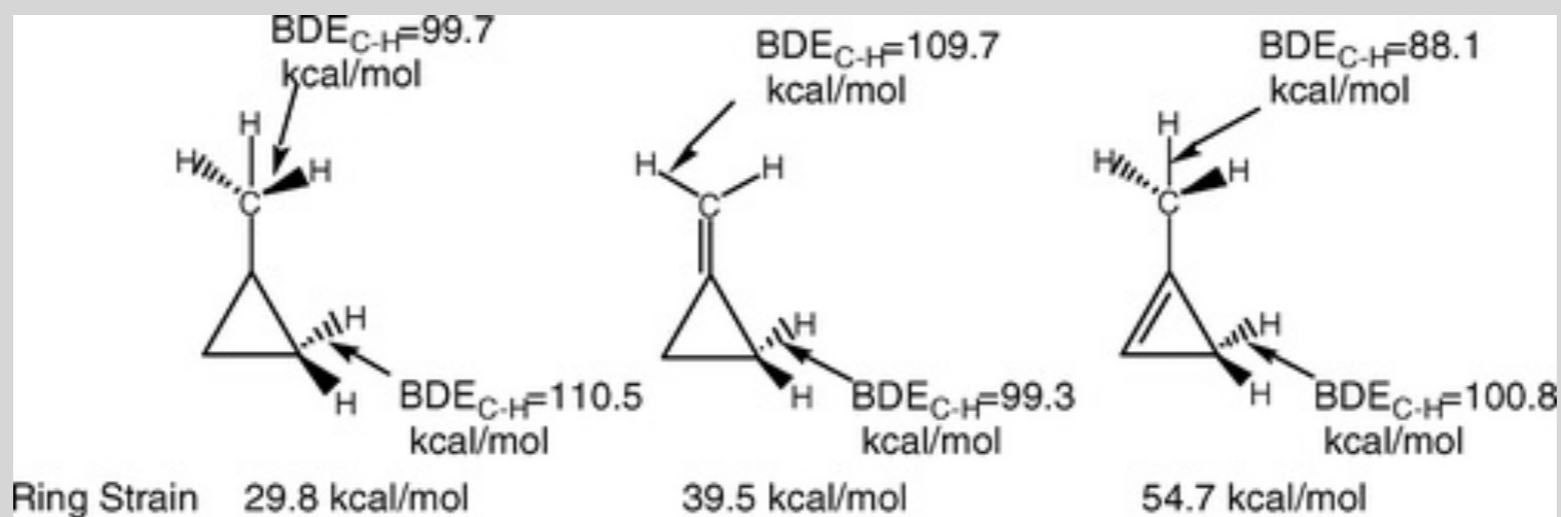
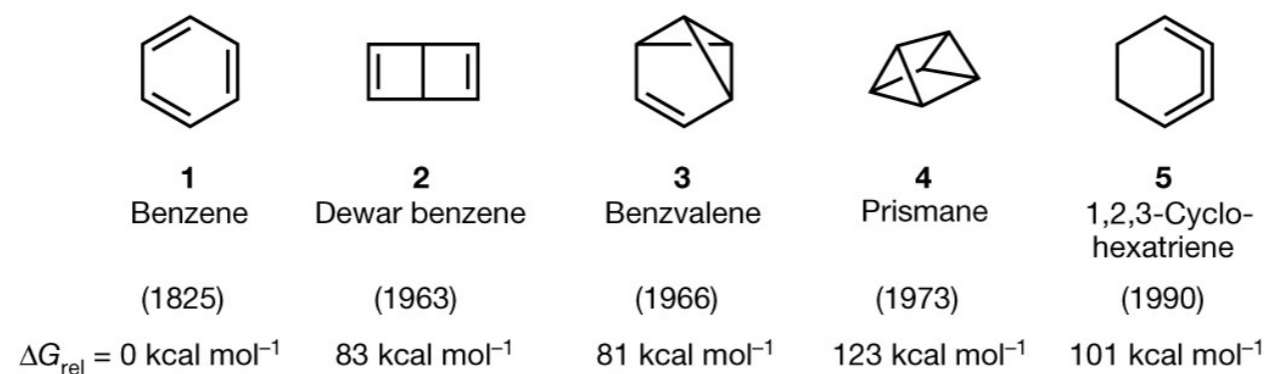
Underexplored
strained intermediates
1,2,3-Cyclohexatrienes



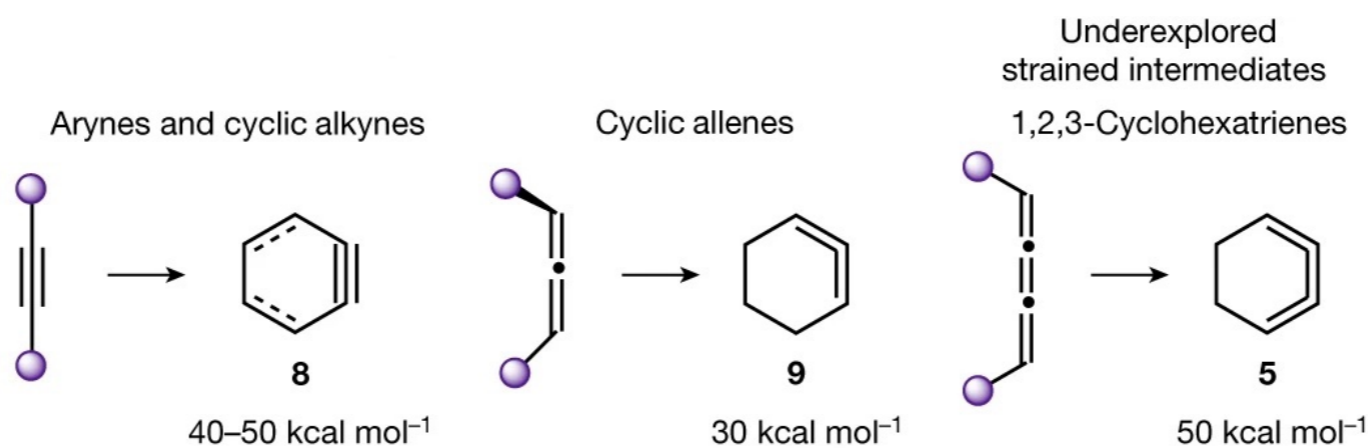
50 kcal mol⁻¹

Computational Studies

LOT B3LYP-D3/6-311+G(d,p)

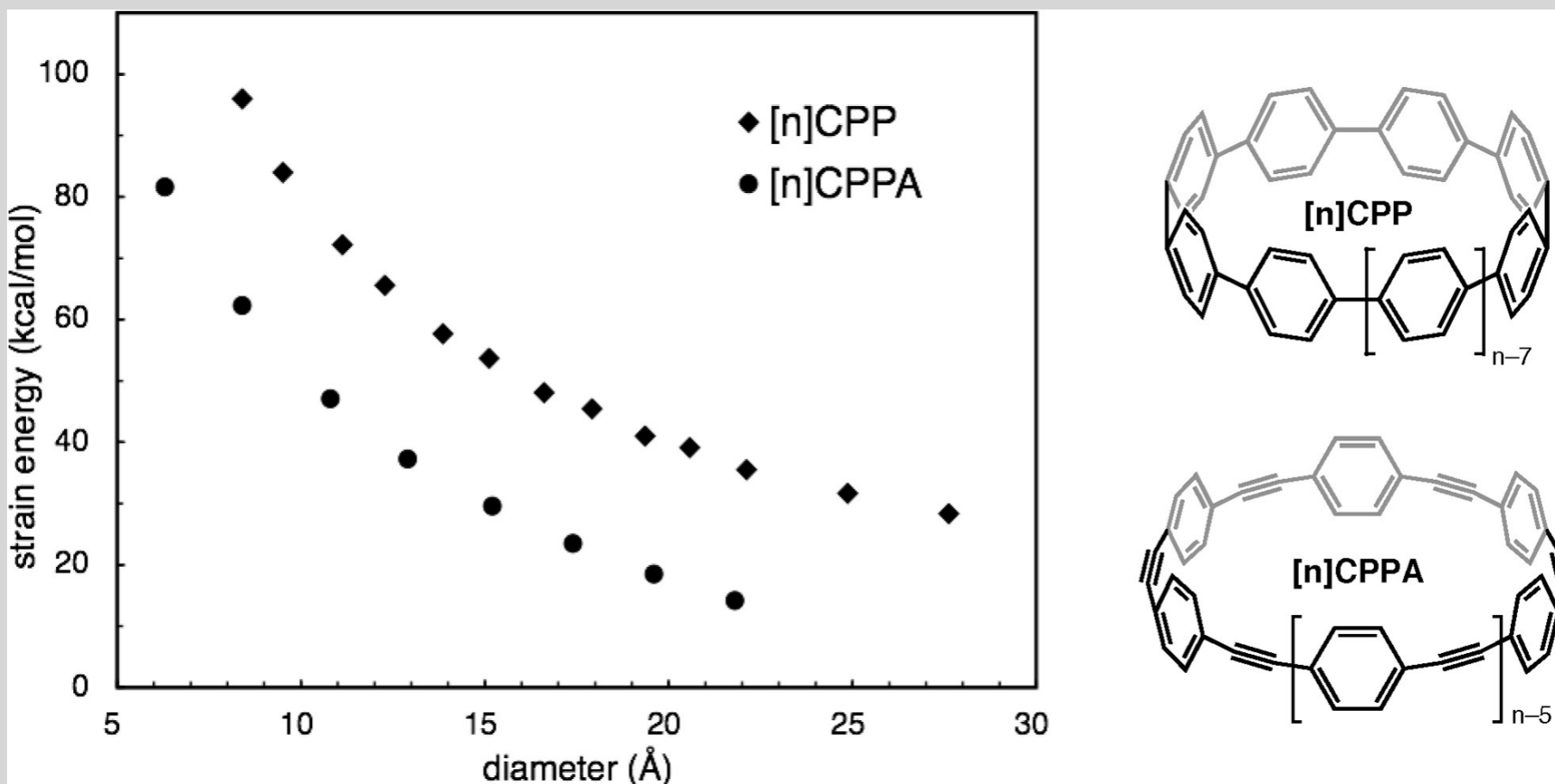
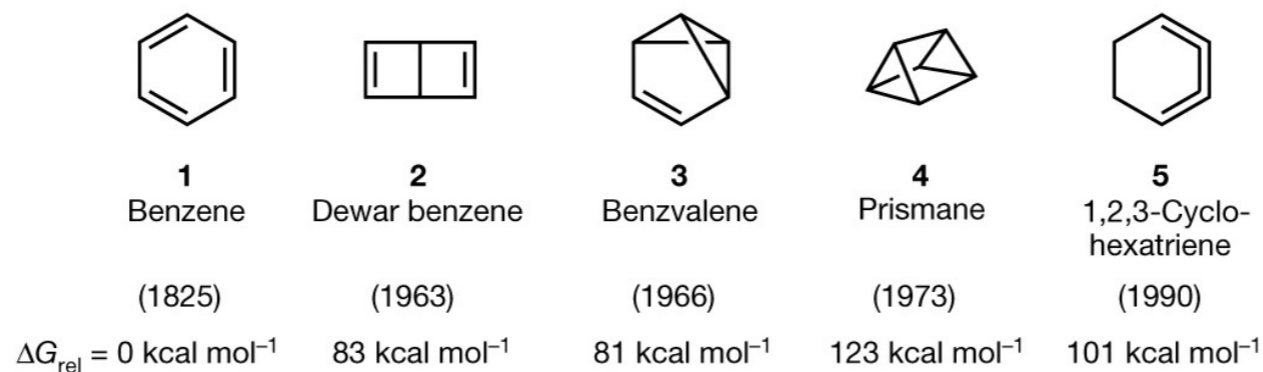


JACS 2004, 126 4444–4452

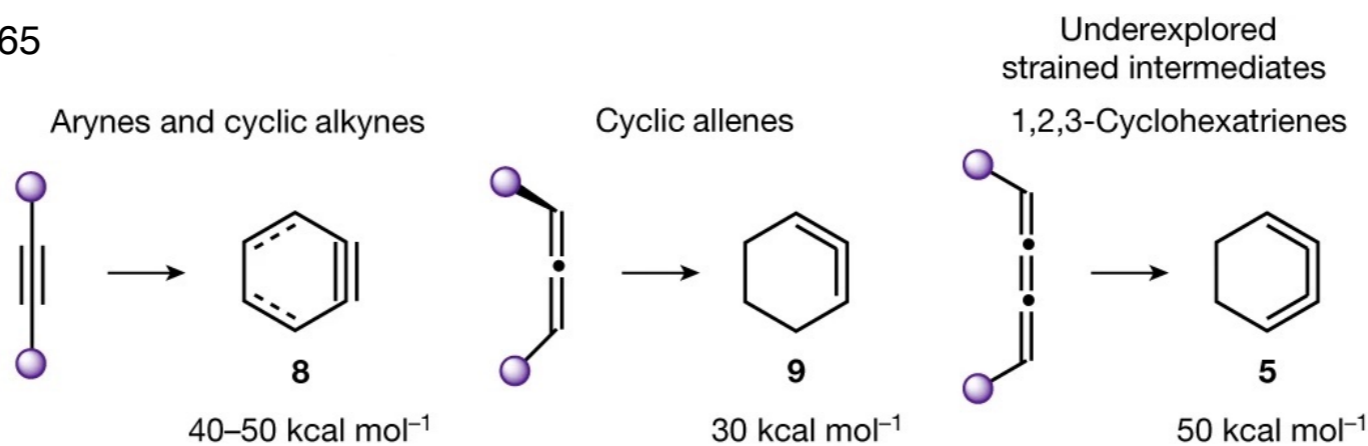


Computational Studies

LOT B3LYP-D3/6-311+G(d,p)

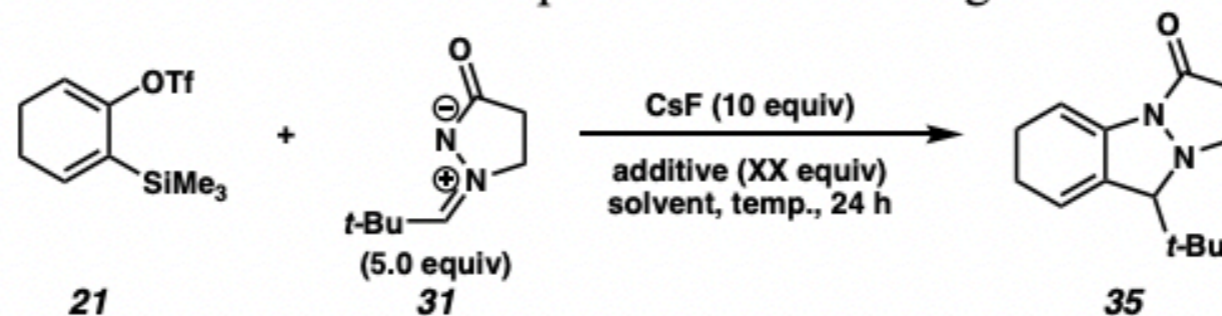


Itami, K. *et al.* *OL* **2010**, 12, 2262–2265



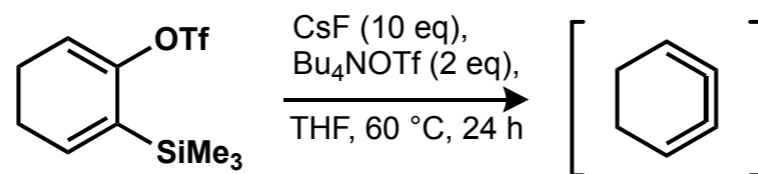
1,2,3-Cyclohexatriene

Supplementary Information– S5

Table S1. Selected Optimization Screening Results

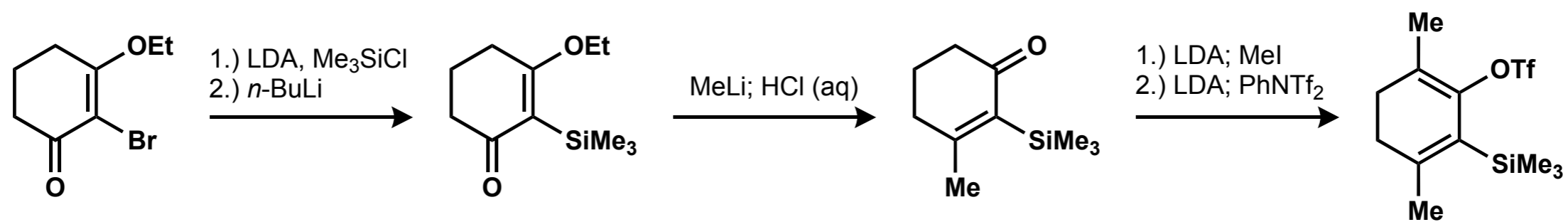
Entry	Conditions	¹ H NMR Yield ^a	Yield BRSM
1	DMSO, 23 °C	26%	26%
2	MeCN, 23 °C	2%	100%
3	Bu ₄ NI (1.0 equiv), MeCN, 60 °C	24%	32%
Solvent Screen			
4	Bu ₄ NI (1.0 equiv), DMSO, 60 °C	24%	24%
5	Bu ₄ NI (1.0 equiv), DMF, 60 °C	28%	28%
6	Bu ₄ NI (1.0 equiv), PhCN, 60 °C	19%	23%
7	Bu ₄ NI (1.0 equiv), Dioxane, 60 °C	14%	61%
8	Bu ₄ NI (1.0 equiv), DME, 60 °C	13%	25%
9	Bu ₄ NI (1.0 equiv), PhH, 60 °C	12%	40%
10	Bu ₄ NI (1.0 equiv), DCE, 60 °C	0%	N/A
11	Bu ₄ NI (1.0 equiv), THF, 60 °C	38%	42%
Fluoride Source Screen			
12	–CsF, +TBAF (5.0 equiv), THF, 23 °C	9%	17%
13	–CsF, +TASF (5.0 equiv), THF, 23 °C	16%	16%
14	CsF (5 equiv), Bu ₄ NOTf (2.0 equiv), THF, 23 °C	44%	46%
Additive Loading Screen			
15	Bu ₄ NOTf (1.0 equiv), THF, 60 °C	42%	76%
16	Bu ₄ NOTf (2.0 equiv), THF, 60 °C	54%	54%
17 ^b	Bu ₄ NOTf (2.0 equiv), THF, 60 °C	36% ^c	40%

Trapping Agent Scope (1/2)

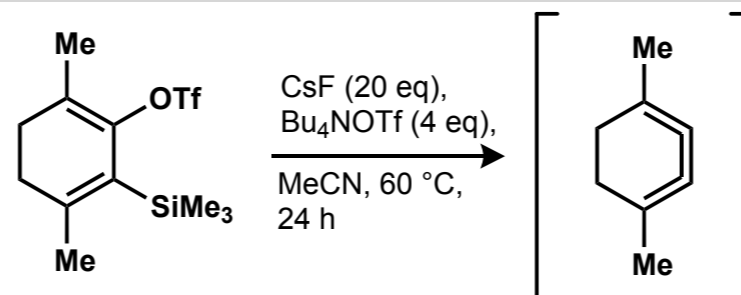


Entry	Trapping agent	Product	Yield	Entry	Trapping agent	Product	Yield
1	23	27	65%	5	31	35	59%
2	24	28	62%	6	32	36	51%
3	25	29	72%	7	33	37	44%
4	26	30	61%	8*	34	38	51%

Trapping Agent Scope (2/3)

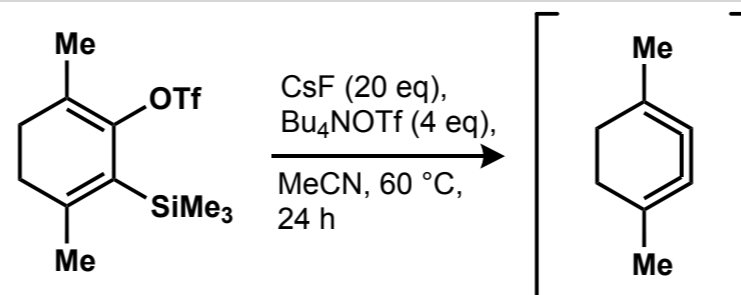


Trapping Agent Scope (3/3)



Entry	Trapping agent	Product and yield	Entry	Trapping agent	Product and yield	Entry	Trapping agent	Product and yield
1 2		 X = O, R ¹ = Me, 24 X = NBoc, R ¹ = H, 26	8		 57 , 79%	13		 69 , 77% (2.7:1 dr)
		X = O, R ¹ = Me, 50 , 80% X = NBoc, R ¹ = H, 51 , 82%		48			61	
3		 52 , 40%	9		 58 , 51%	14 15		 X = O, R ⁶ = NO ₂ , 70 , 83% X = S, R ⁶ = H, 71 , 81%
	45			49			Nu = NaO, R ⁶ = NO ₂ , 34 Nu = SH, R ⁶ = H, 62	
4 5		 R ² = <i>t</i> -Bu, 53 , 86% R ² = Ph, 54 , 87%	10		 66 , 68%	16		 72 , 58%
	R ² = <i>t</i> -Bu, 31 R ² = Ph, 32			59			63	
6 7		 R ³ = Ph, 55 , 74% R ³ = <i>t</i> -Bu, 56 , 75%	11* 12*		 R ⁴ = R ⁵ = OEt, 67 , 47% R ⁴ = NBoc ₂ , R ⁵ = CO ₂ Me, 68 , 48%	17 18		 R ⁷ = Et, R ⁸ = CN, 73 , 47% R ⁷ = Me, R ⁸ = CO ₂ Me, 74 , 82%
	R ³ = Ph, 46 R ³ = <i>t</i> -Bu, 47			60			R ⁷ = Et, R ⁸ = CN, 64 R ⁷ = Me, R ⁸ = CO ₂ Me, 65	

Trapping Agent Scope (3/3)

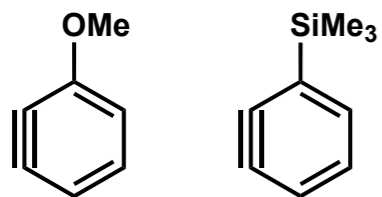


Entry	Trapping agent	Product and yield	Entry	Trapping agent	Product and yield	Entry	Trapping agent	Product and yield
1 2			8			13		

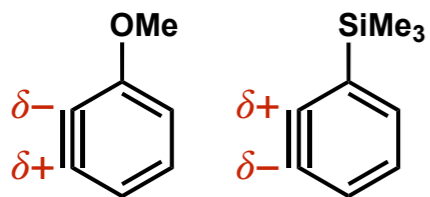
switching the solvent to acetonitrile and increasing the loading of Bu_4NOTf facilitated full conversion and good yields in trappings of **43**. More generally, we observed that trappings of disubstituted triene **43** proceeded in higher yields and with broader scope compared with trappings of the parent compound, 1,2,3-cyclohexatriene (**5**), despite **43** bearing an increased steric profile. The results indicate the engagement of disubstituted triene **43** in (a) (4 + 2)

4 5			10			16		
	$\text{R}^2 = t\text{-Bu}$, 31 $\text{R}^2 = \text{Ph}$, 32	$\text{R}^2 = t\text{-Bu}$, 53 , 86% $\text{R}^2 = \text{Ph}$, 54 , 87%	59	66 , 68%	63	72 , 58%		
6 7			11* 12*			17 18		
	$\text{R}^3 = \text{Ph}$, 46 $\text{R}^3 = t\text{-Bu}$, 47	$\text{R}^3 = \text{Ph}$, 55 , 74% $\text{R}^3 = t\text{-Bu}$, 56 , 75%	$\text{R}^4 = \text{R}^5 = \text{OEt}$, 33 $\text{R}^4 = \text{NBoc}_2$, $\text{R}^5 = \text{CO}_2\text{Me}$, 60	$\text{R}^4 = \text{R}^5 = \text{OEt}$, 67 , 47% $\text{R}^4 = \text{NBoc}_2$, $\text{R}^5 = \text{CO}_2\text{Me}$, 68 , 48%	$\text{R}^7 = \text{Et}$, $\text{R}^8 = \text{CN}$, 64 $\text{R}^7 = \text{Me}$, $\text{R}^8 = \text{CO}_2\text{Me}$, 65	$\text{R}^7 = \text{Et}$, $\text{R}^8 = \text{CN}$, 73 , 47% $\text{R}^7 = \text{Me}$, $\text{R}^8 = \text{CO}_2\text{Me}$, 74 , 82%		

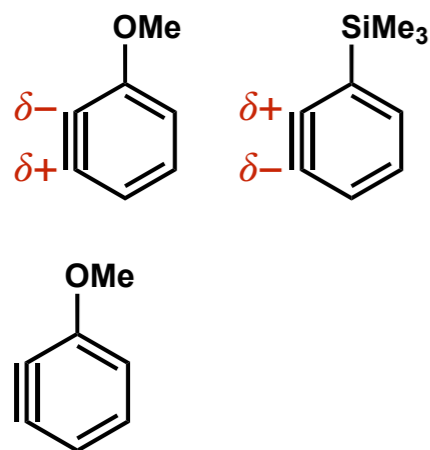
Selectivity of Aryne Chemistry: an Analogy



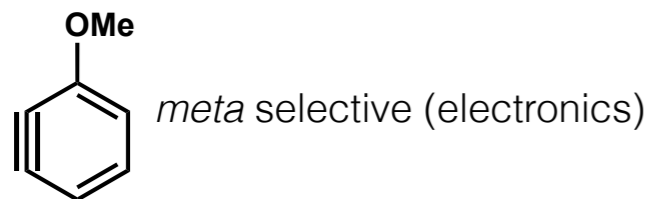
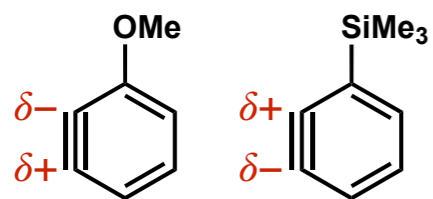
Selectivity of Aryne Chemistry: an Analogy



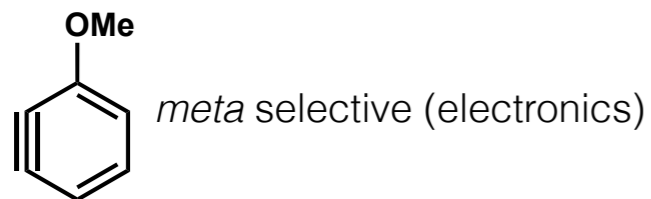
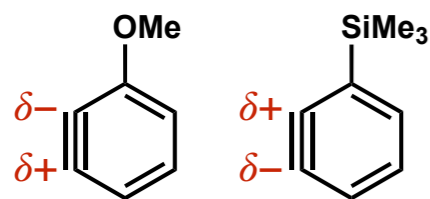
Selectivity of Aryne Chemistry: an Analogy



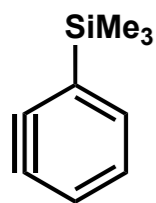
Selectivity of Aryne Chemistry: an Analogy



Selectivity of Aryne Chemistry: an Analogy

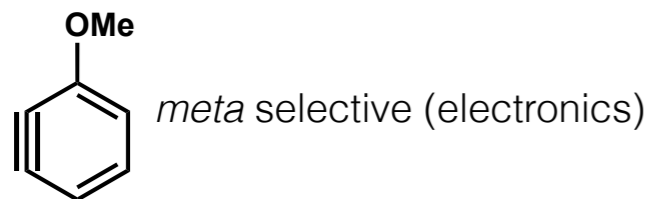
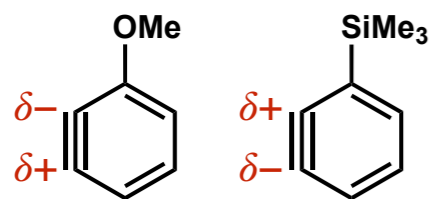


But how about?

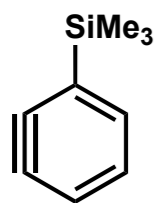


Monosubstituted Triene Case Study

Selectivity of Aryne Chemistry: an Analogy



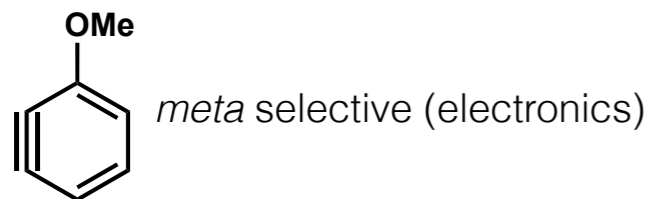
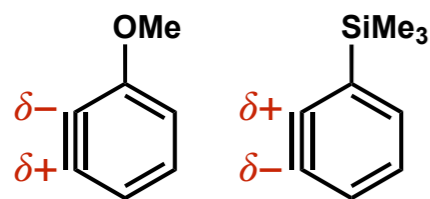
But how about?



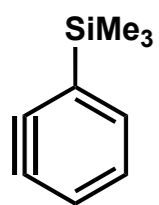
entry	trapping agent	product(s)	ratio (yield ^b)
1	<chem>H2N-Bn</chem>	 <chem>C[Si](C)Oc1ccc(NC(=O)Cc2ccccc2)cc1</chem> <i>ortho</i>	1:2 (81% yield)
2	<chem>H2N-C6H5</chem>	 <chem>C[Si](C)Oc1cccc(NC(=O)Cc2ccccc2)c1</chem> <i>meta</i>	2:1 (99% yield)
3	<chem>HS-C6H4-Me</chem>	 <chem>C[Si](C)Oc1cccc(Sc2ccc(C)cc2)c1</chem> <i>meta</i>	4:1 (91% yield)
4	<chem>HS-C6H3(CO2Me)</chem>	 <chem>C[Si](C)Oc1cccc(Sc2ccc(C(=O)OC)cc2)c1</chem> <i>meta</i>	10:1 (77% yield)
5	<chem>H2N-C6H10=O</chem>	 <chem>C[Si](C)Oc1cccc(Sc2c(N)ccc2=O)c1</chem> <i>meta</i>	(52% yield)
6	<chem>HO-C6H4-t-Bu</chem>	 <chem>C[Si](C)Oc1cccc(OC(=O)c2ccc(C(C)(C)C)cc2)c1</chem> <i>ortho</i>	1:3 (53% yield)

Monosubstituted Triene Case Study

Selectivity of Aryne Chemistry: an Analogy

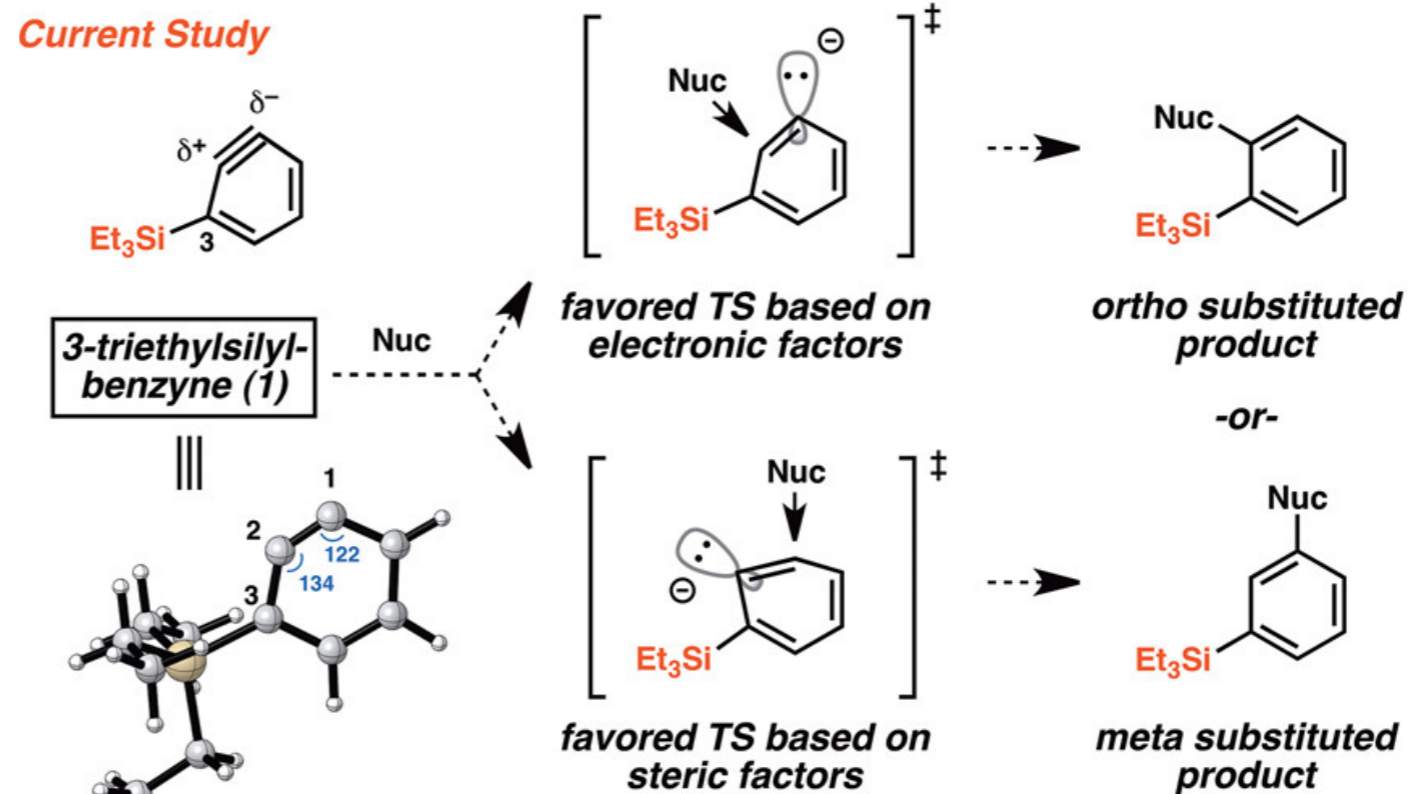
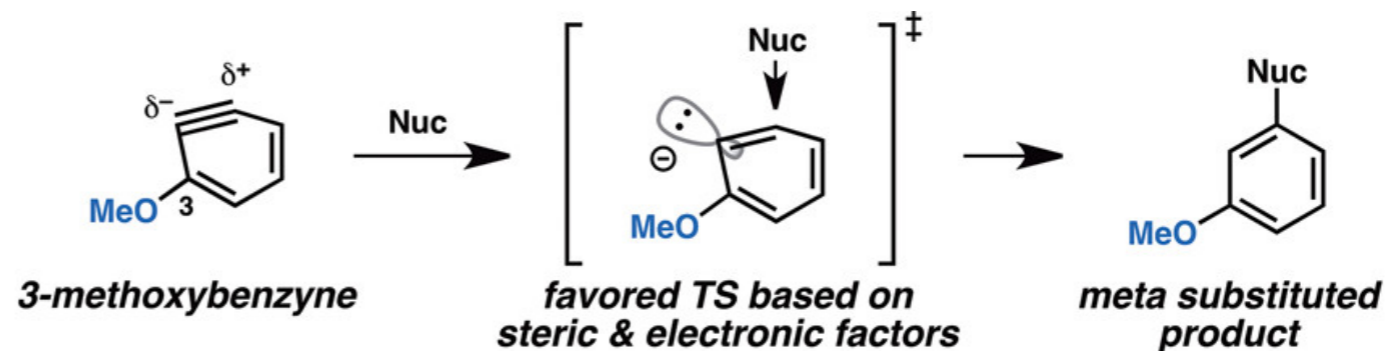


But how about?



Introducing...

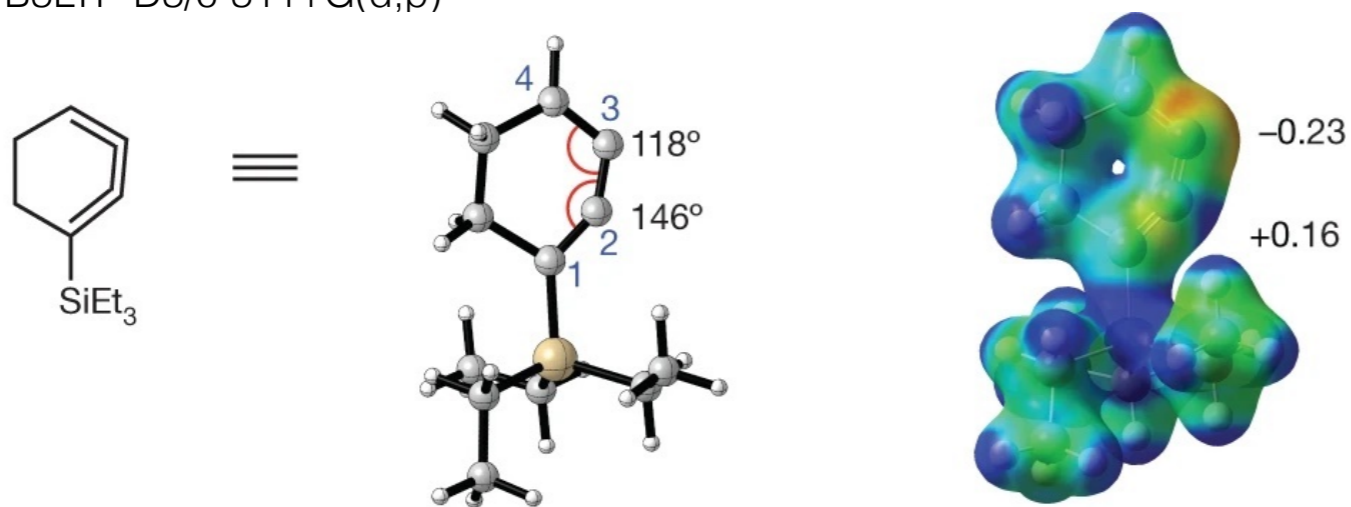
The aryne distortion model



LOT B3LYP/6-311+G(d,p)

Monosubstituted Triene Case Study

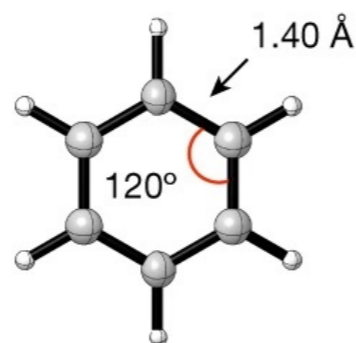
LOT B3LYP-D3/6-311+G(d,p)



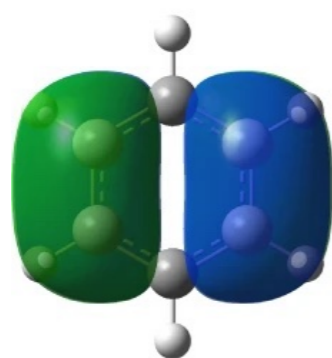
Comparison...



\equiv

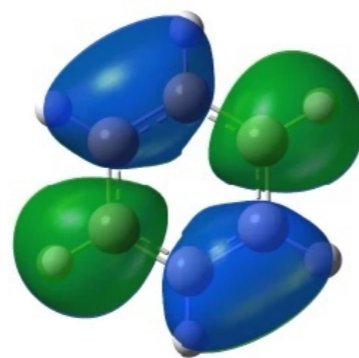


HOMO

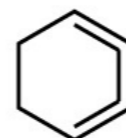


(-7.1 eV)

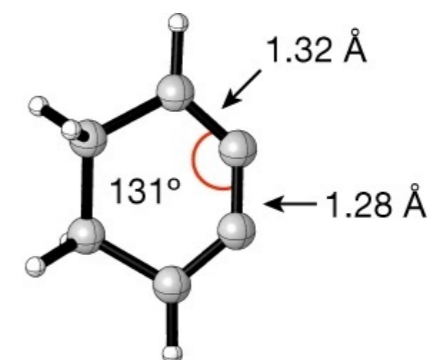
LUMO



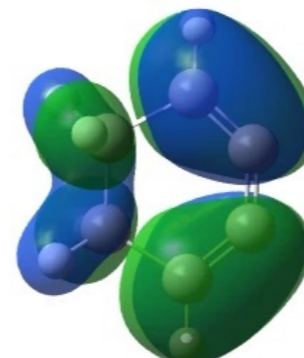
(-0.5 eV)



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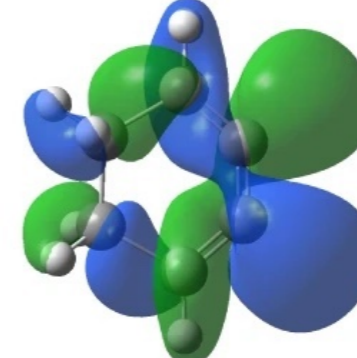


HOMO



(-6.2 eV)

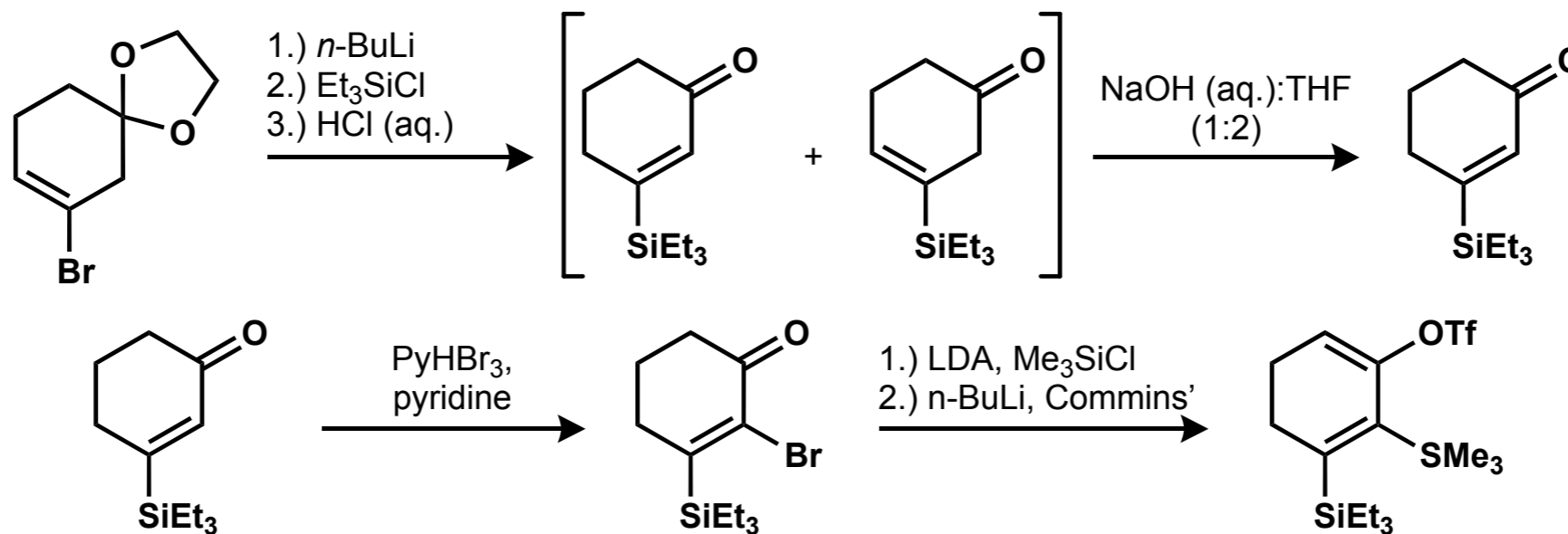
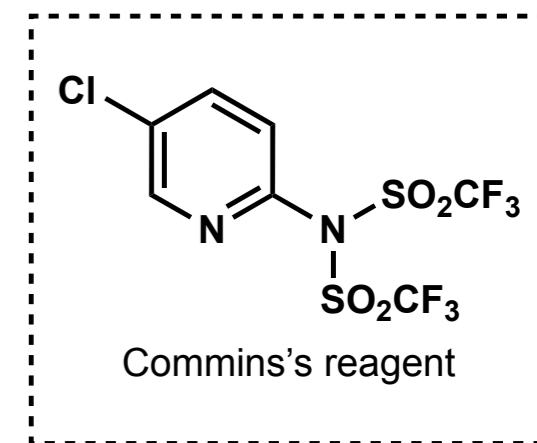
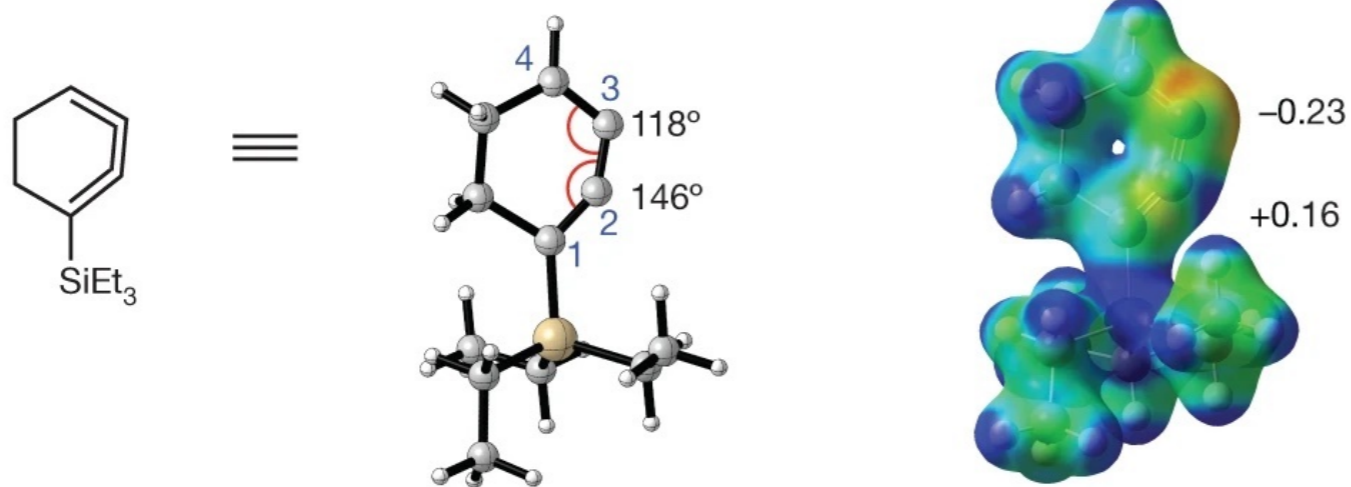
LUMO



(-2.1 eV)

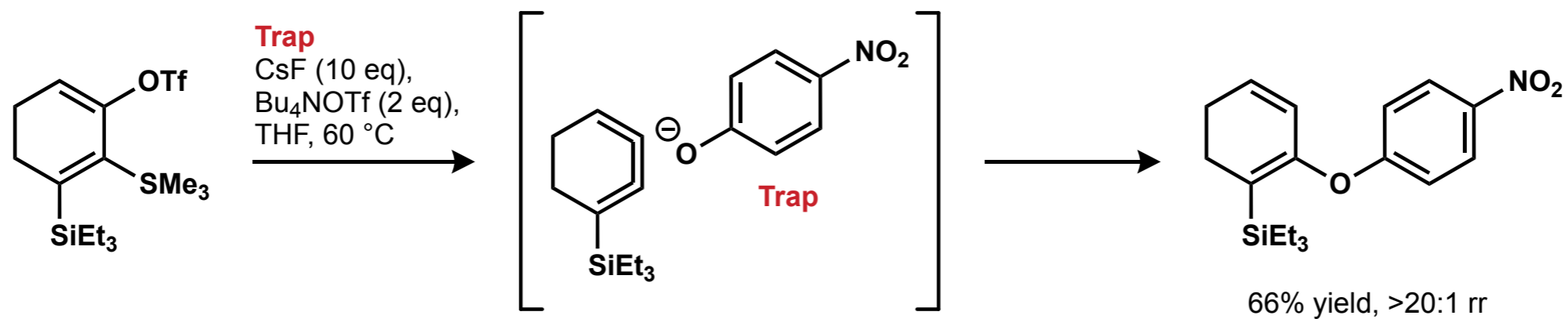
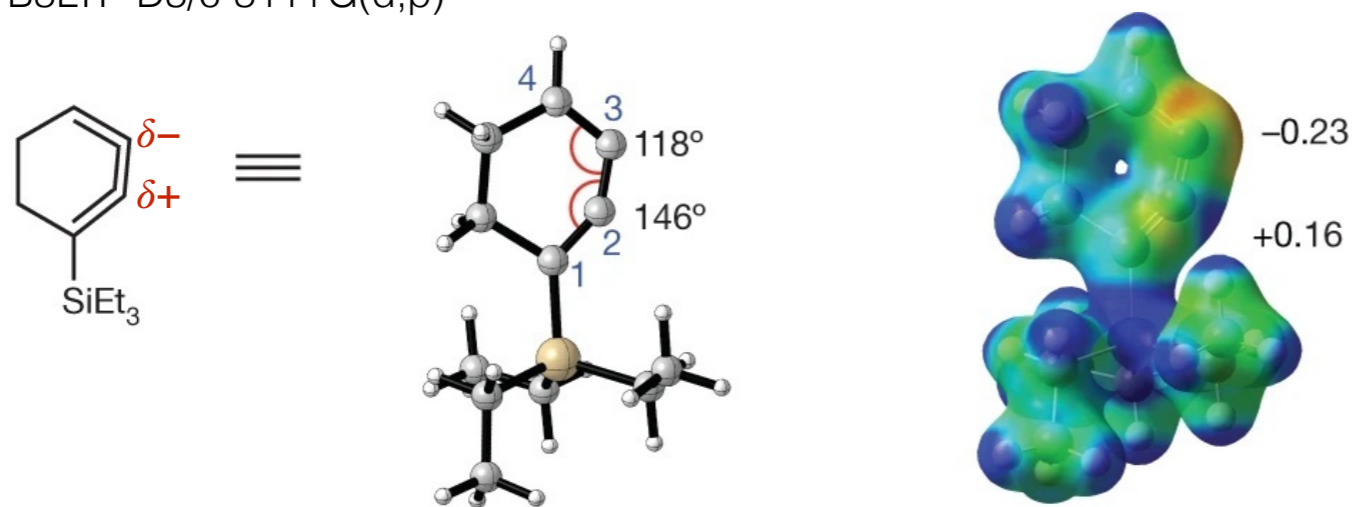
Monosubstituted Triene Case Study

LOT B3LYP-D3/6-311+G(d,p)



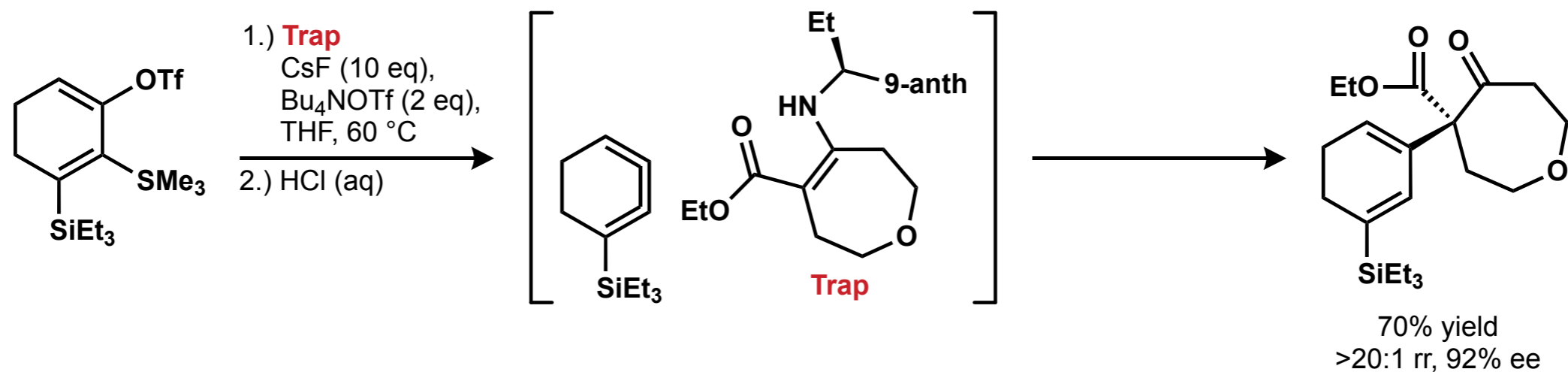
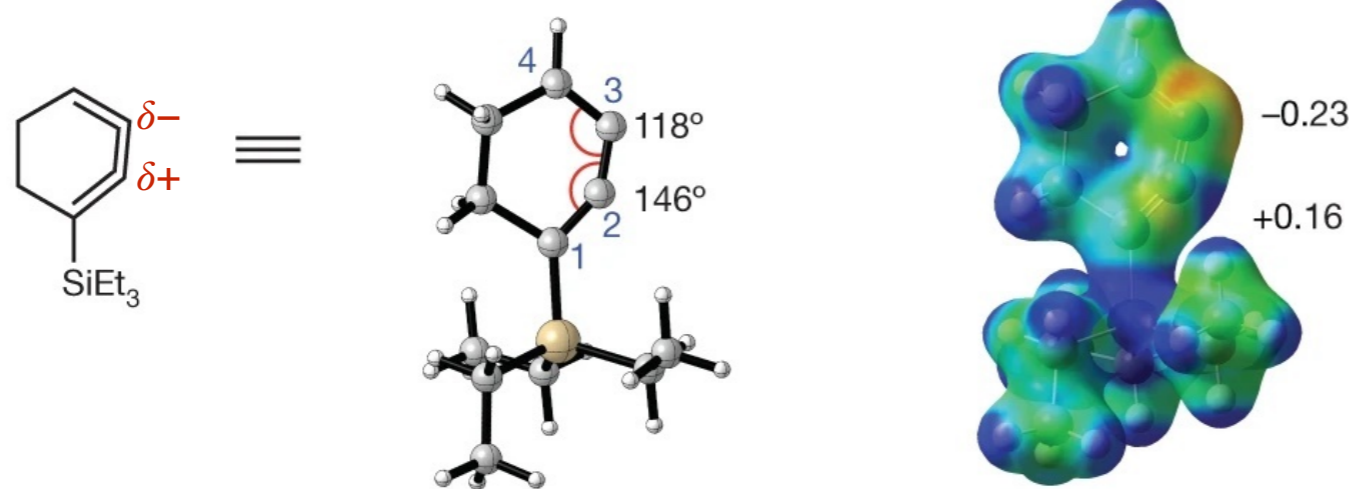
Monosubstituted Triene Case Study and Application 23

LOT B3LYP-D3/6-311+G(d,p)



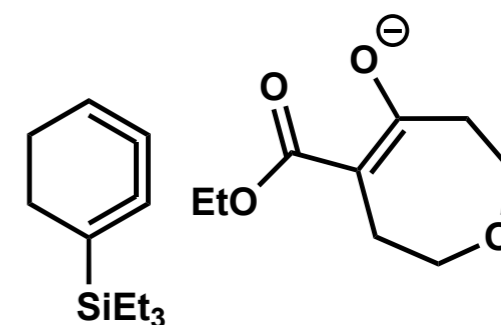
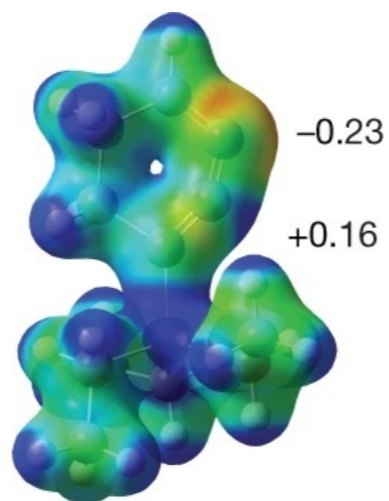
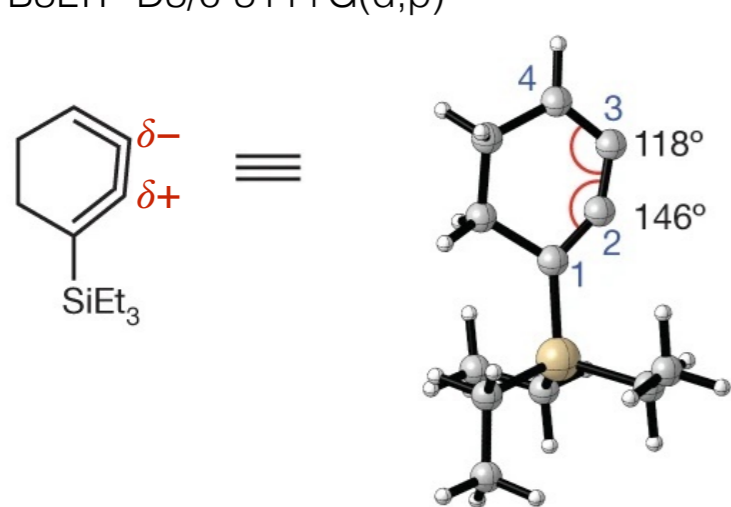
Monosubstituted Triene Case Study and Application 24

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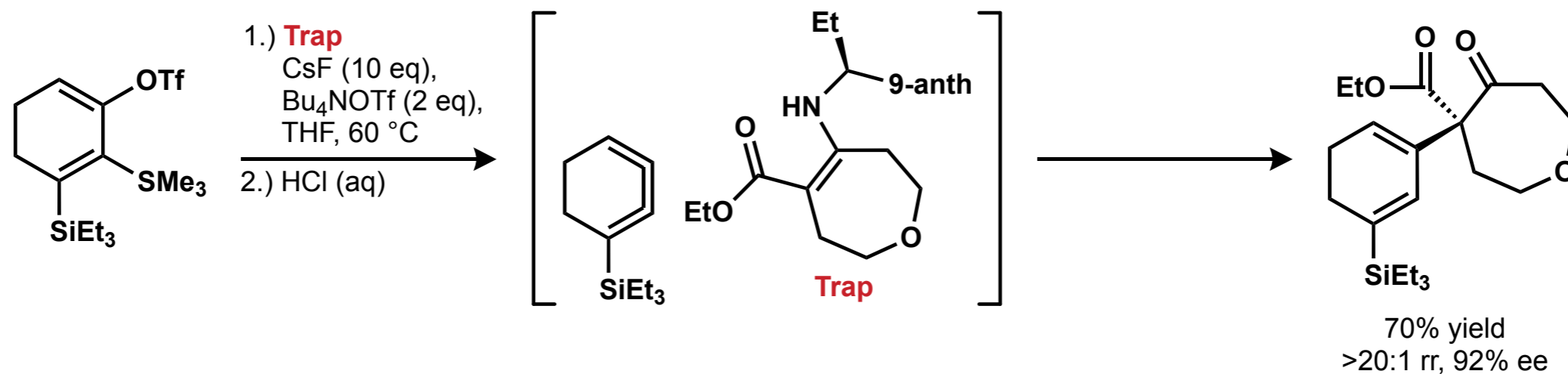


Monosubstituted Triene Case Study and Application 25

LOT B3LYP-D3/6-311+G(d,p)

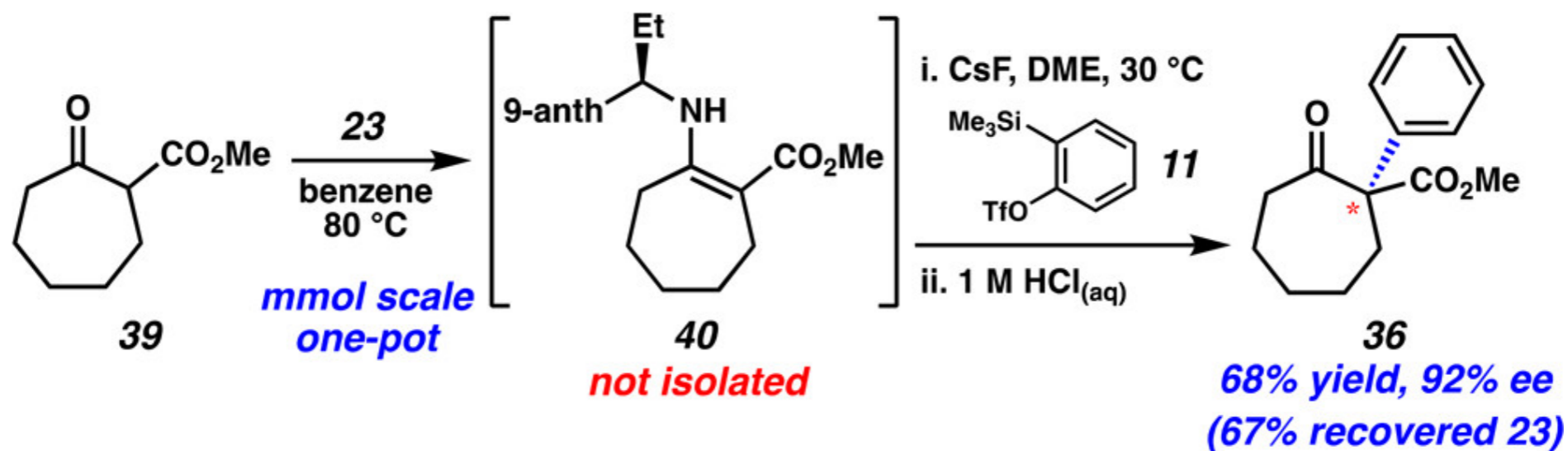
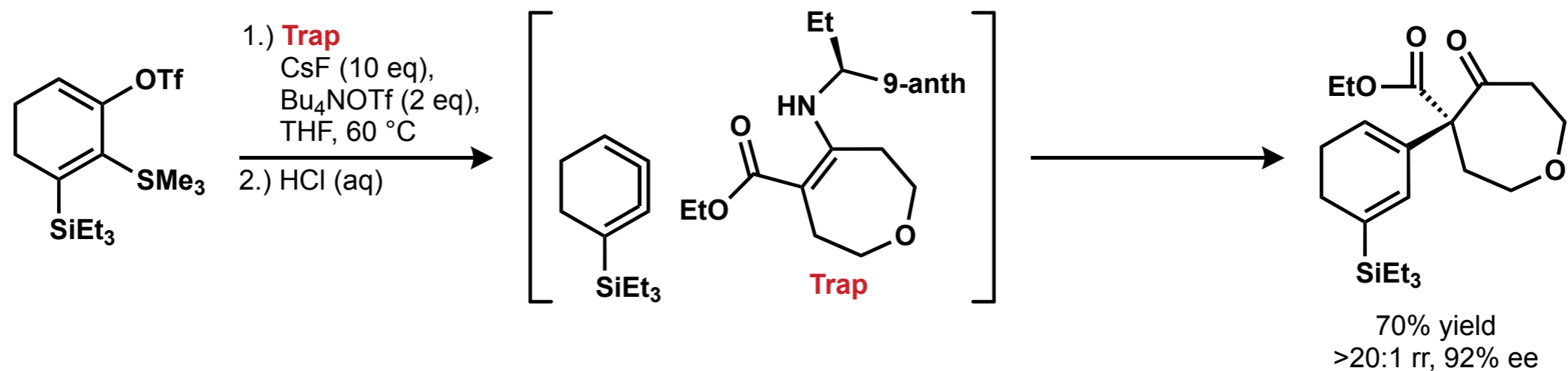
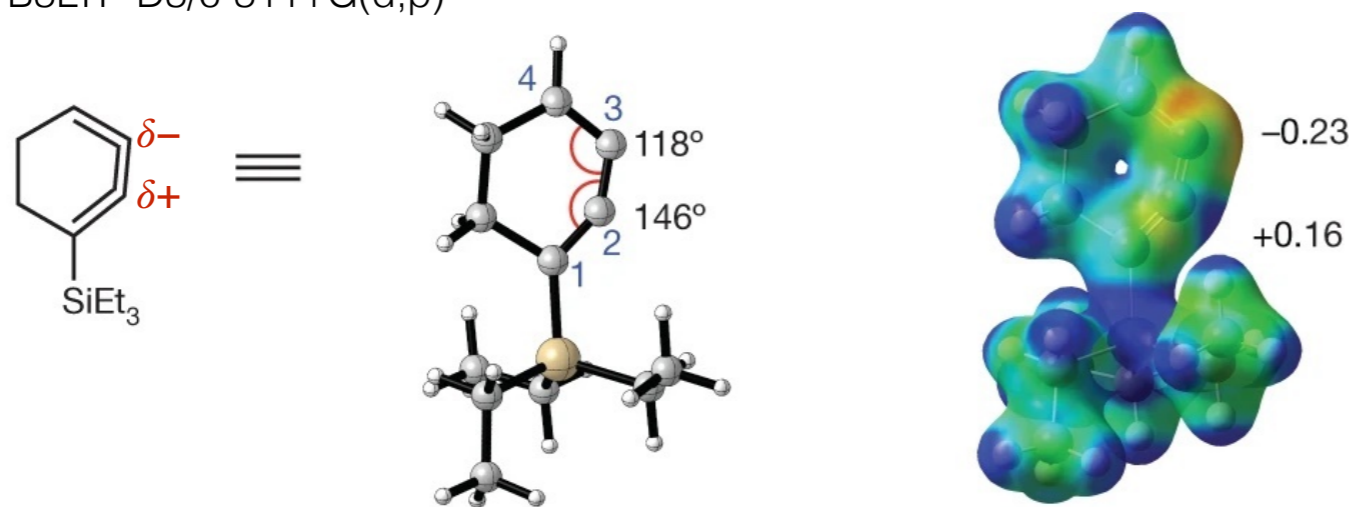


Direct from the ketoester?



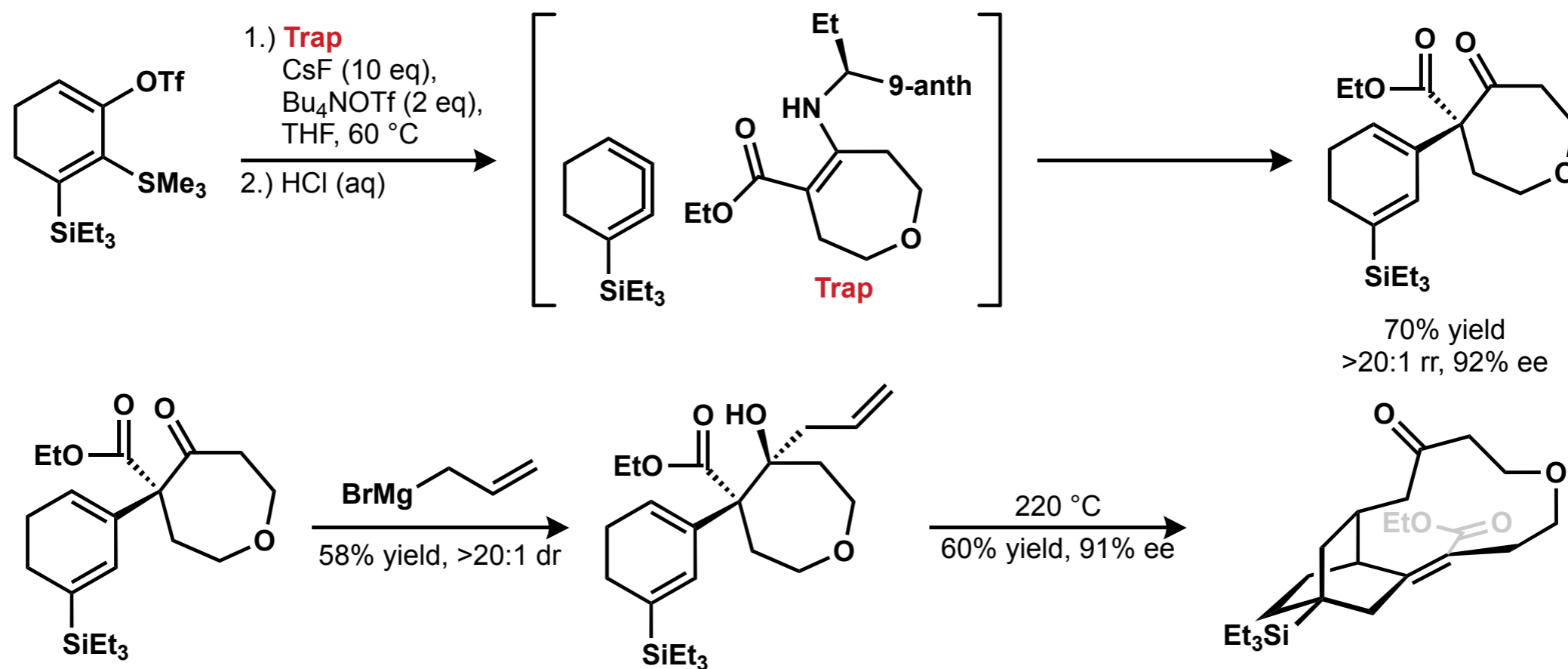
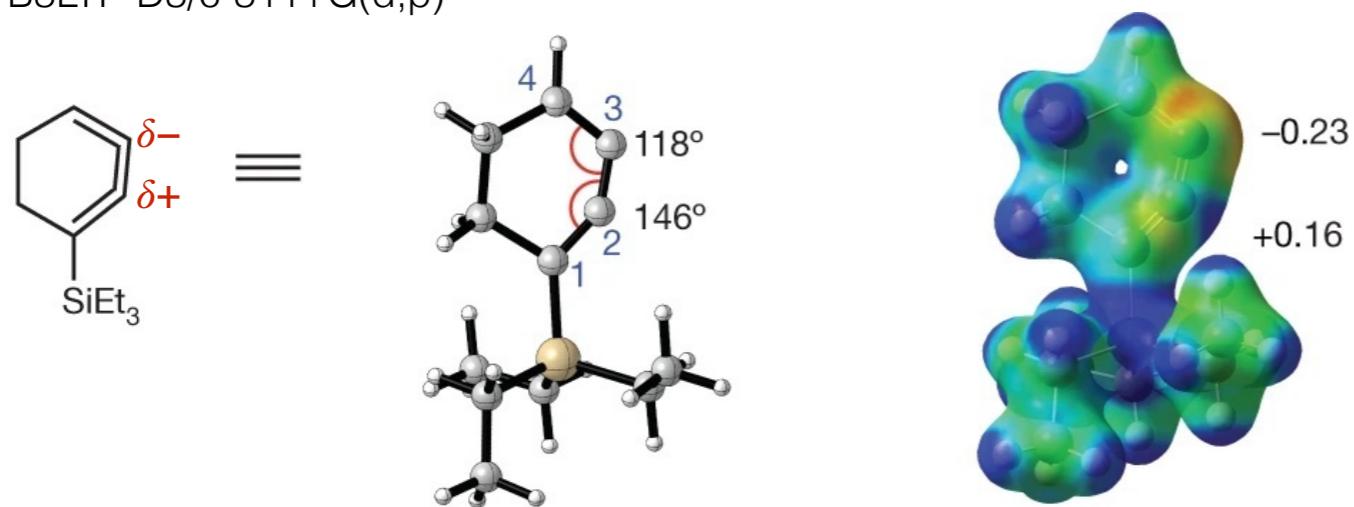
Monosubstituted Triene Case Study and Application 26

LOT B3LYP-D3/6-311+G(d,p)



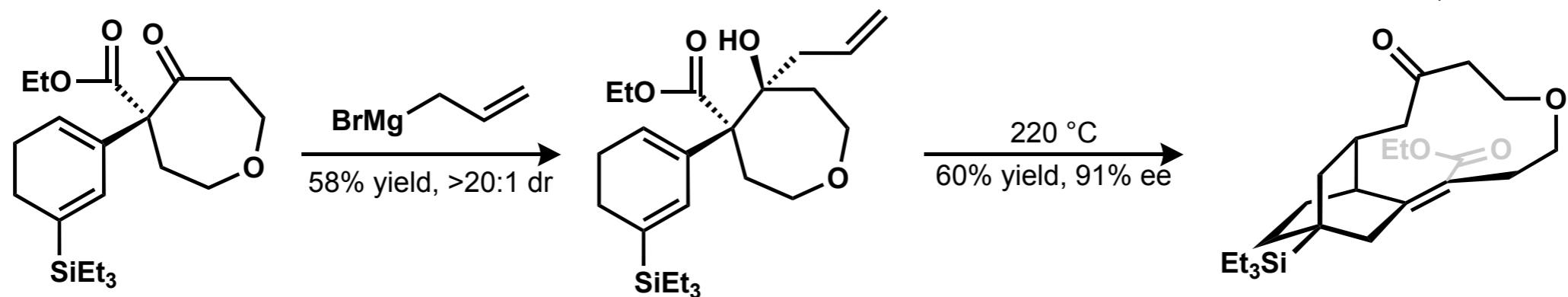
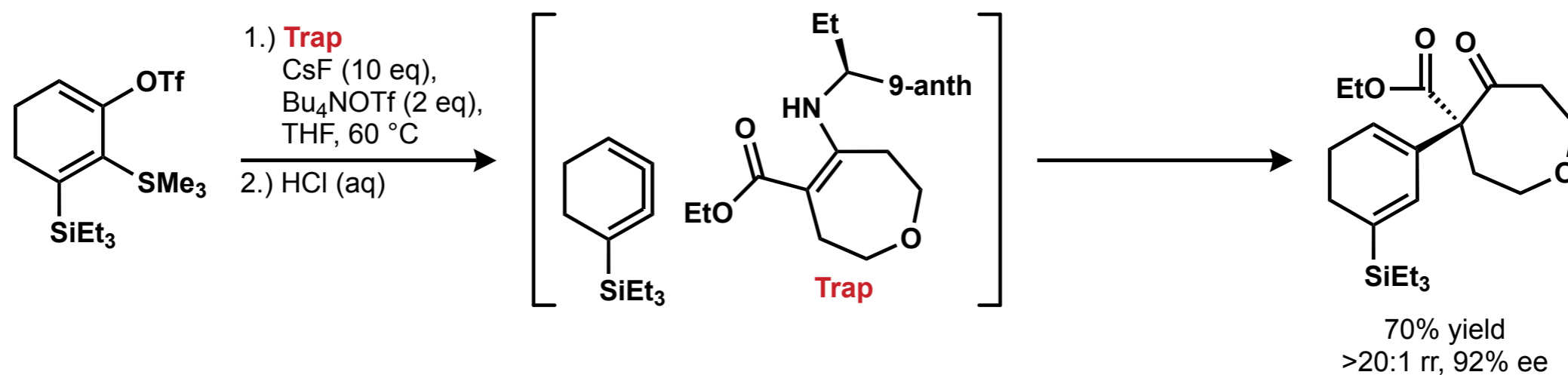
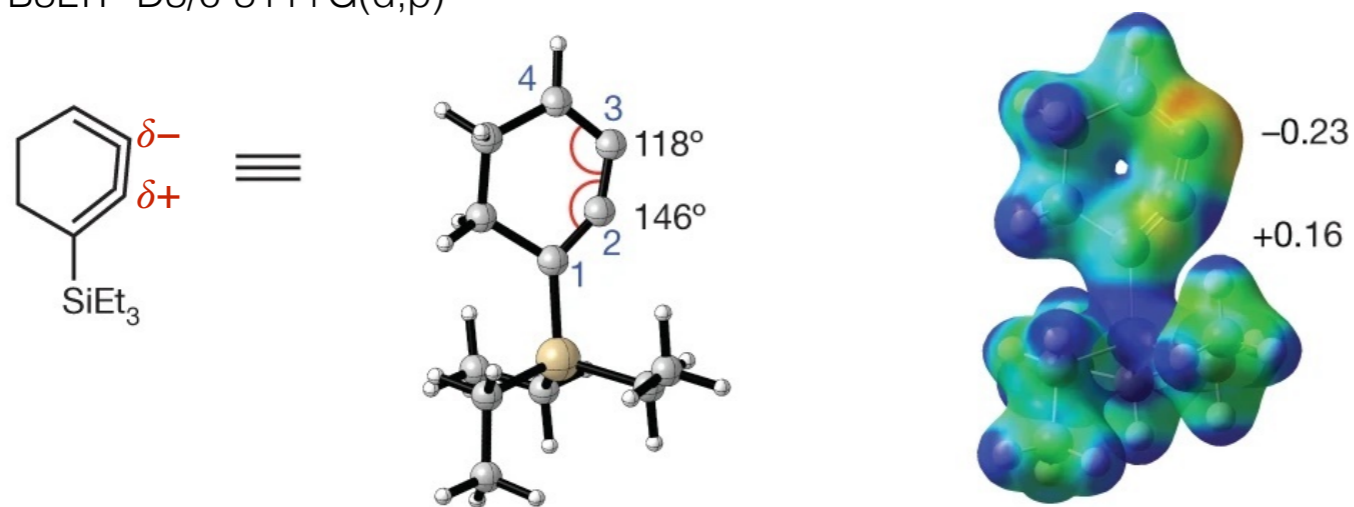
Monosubstituted Triene Case Study and Application 27

LOT B3LYP-D3/6-311+G(d,p)



Monosubstituted Triene Case Study and Application 28

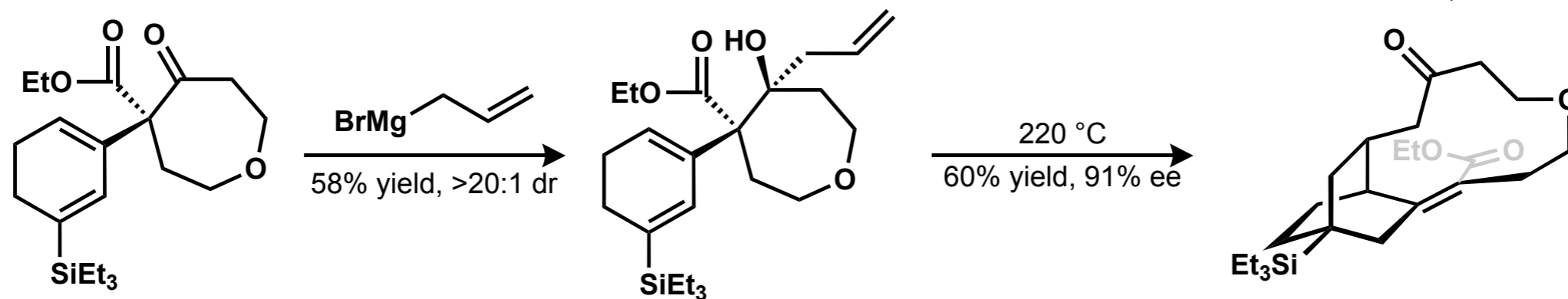
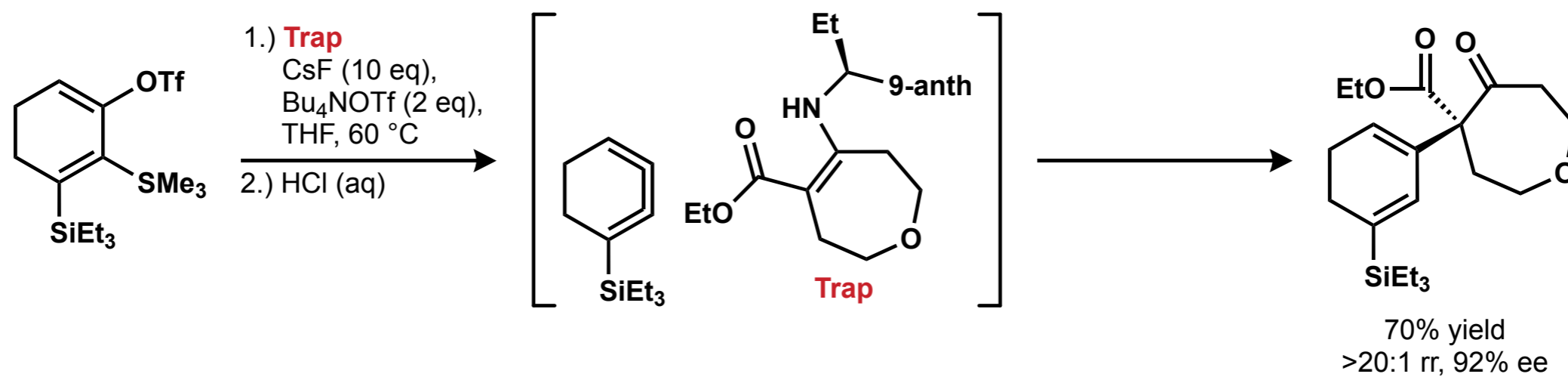
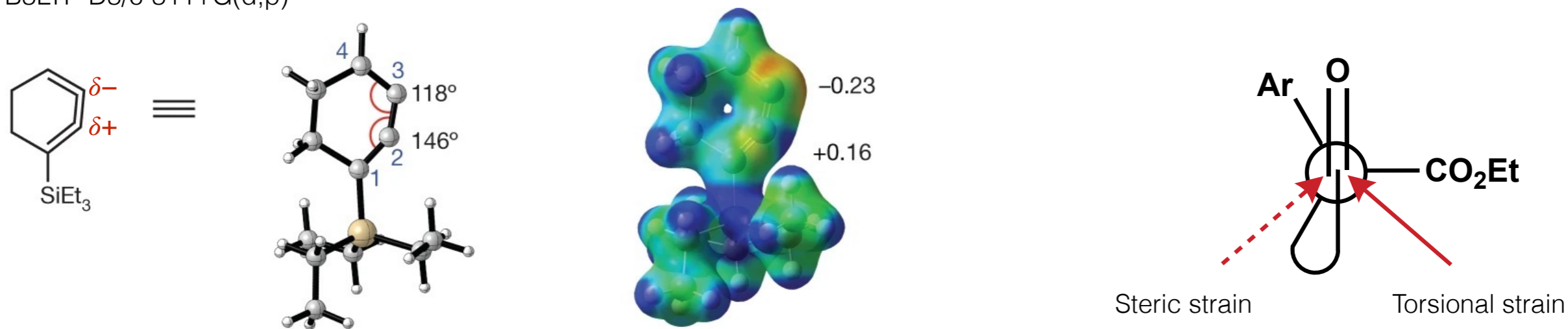
LOT B3LYP-D3/6-311+G(d,p)



Felkin-Anh??

Monosubstituted Triene Case Study and Application 29

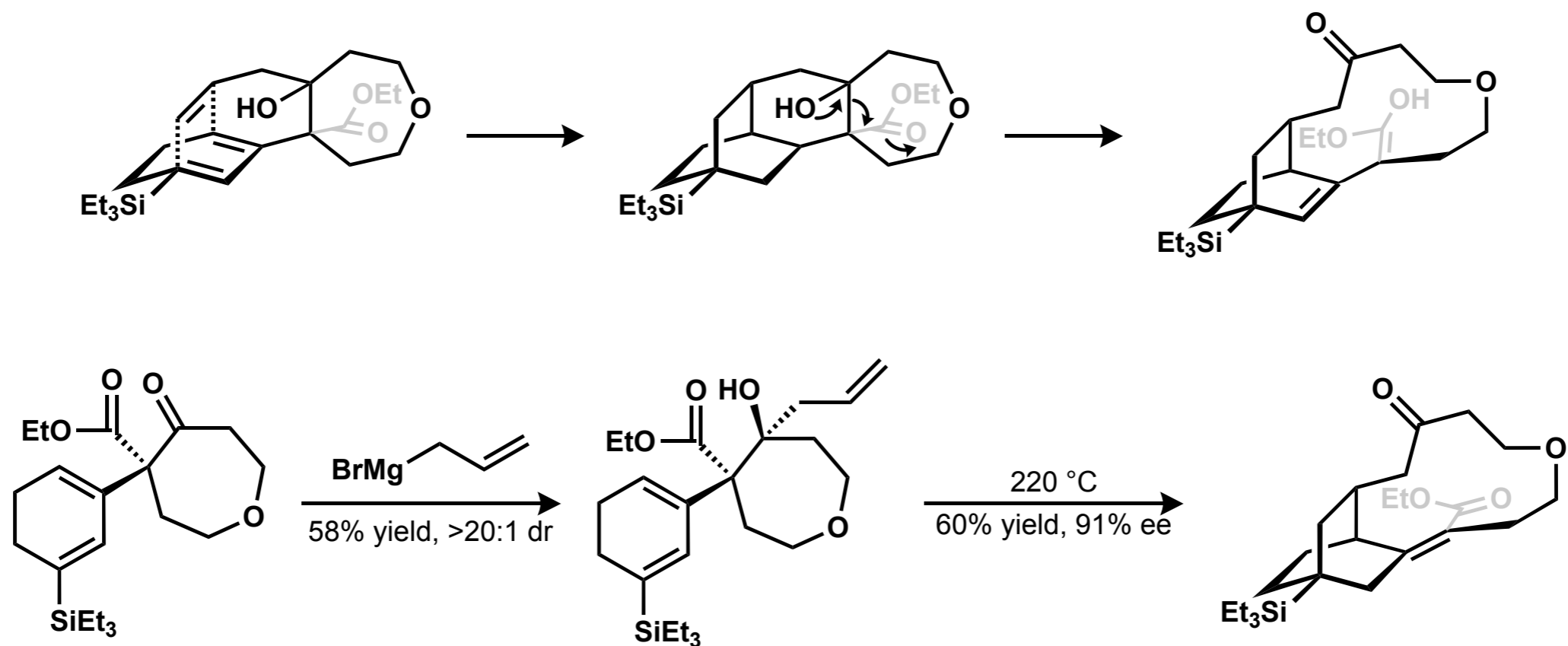
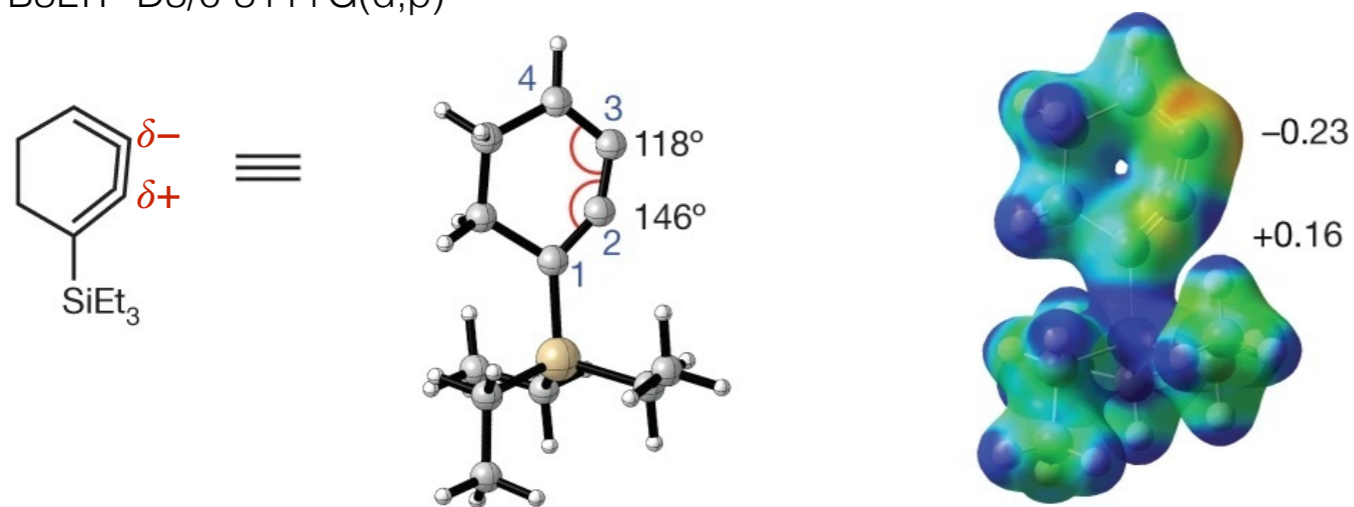
LOT B3LYP-D3/6-311+G(d,p)



Felkin-Anh??

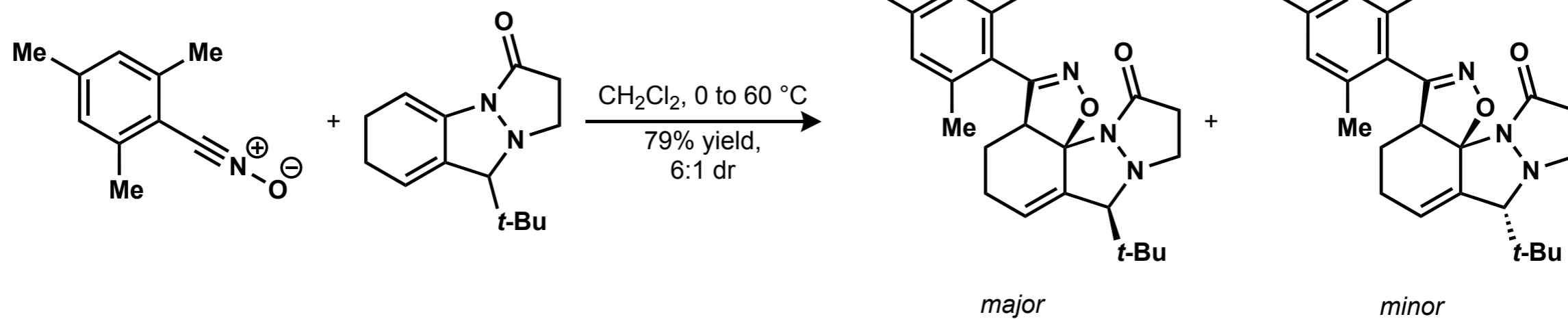
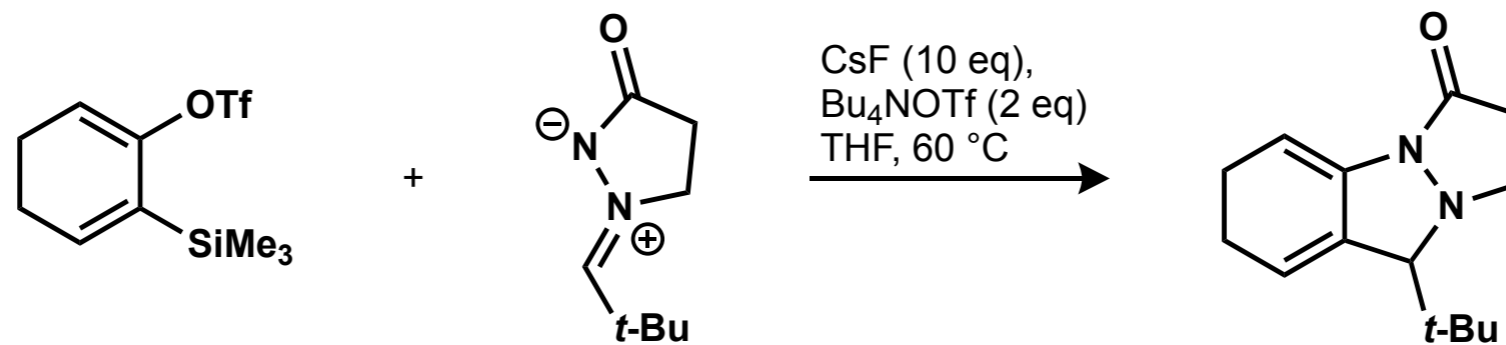
Monosubstituted Triene Case Study and Application 30

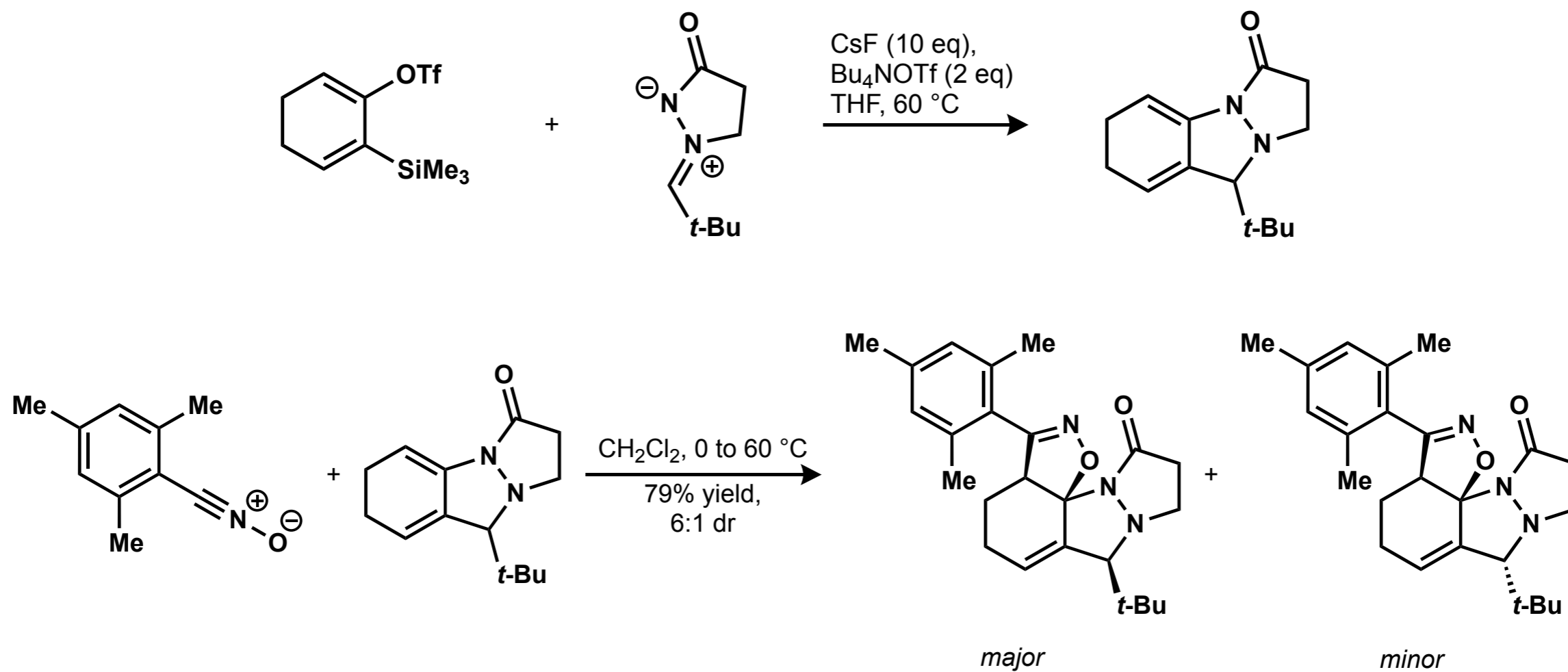
LOT B3LYP-D3/6-311+G(d,p)



Chiral auxiliary *beta*-ketoesters Garg, N.; Houk, K. N. *et al. JACS* **2018**, *140*(24), 7605–7610

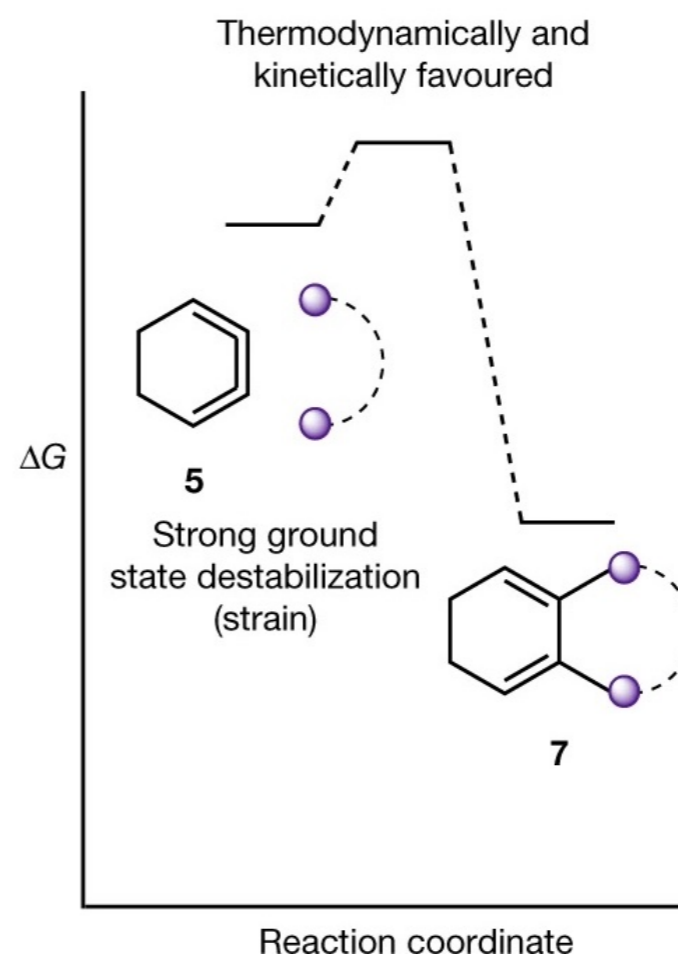
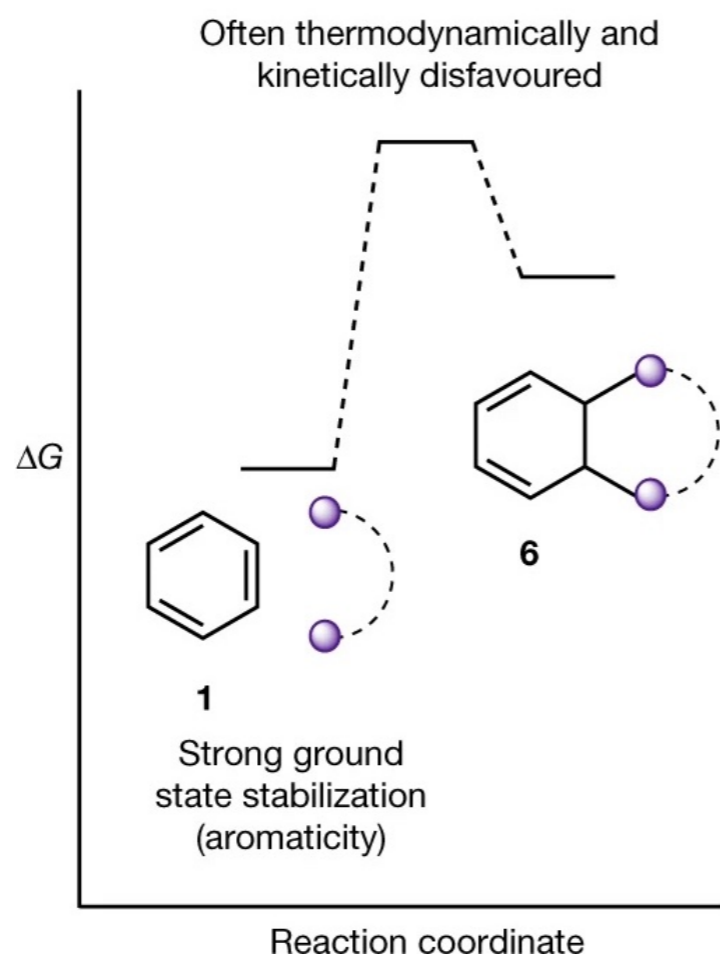
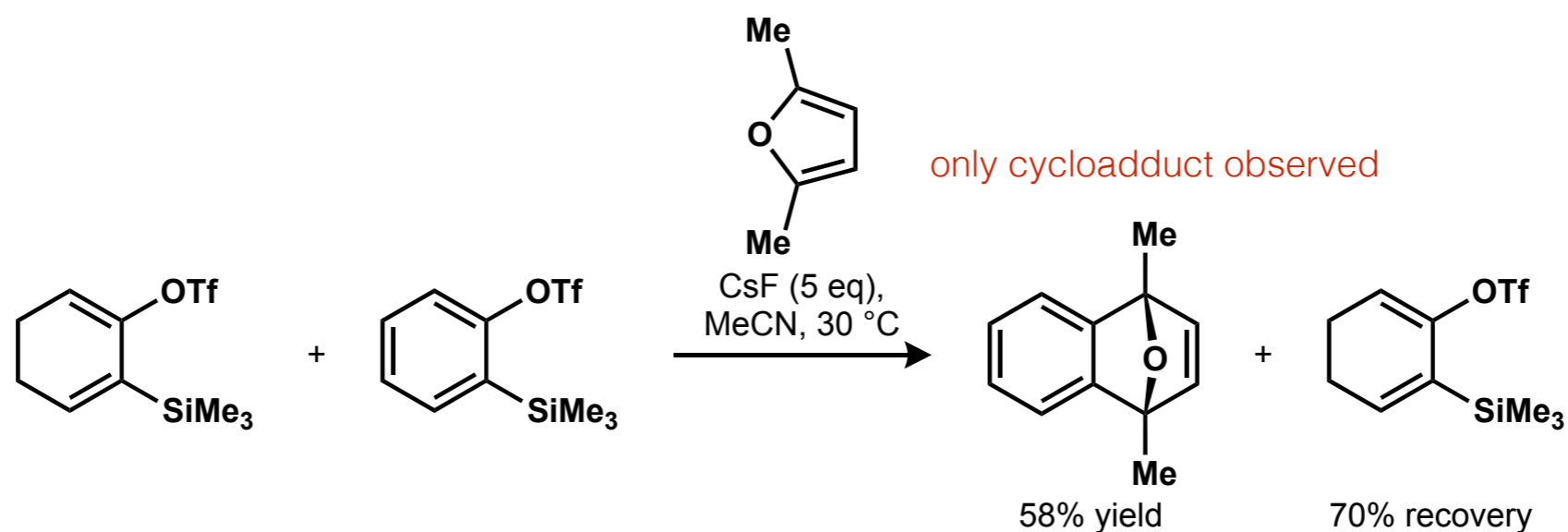
C–C cleavage Stoltz, B. M. *et al. JACS* **2005**, *127*(15), 5340–5341



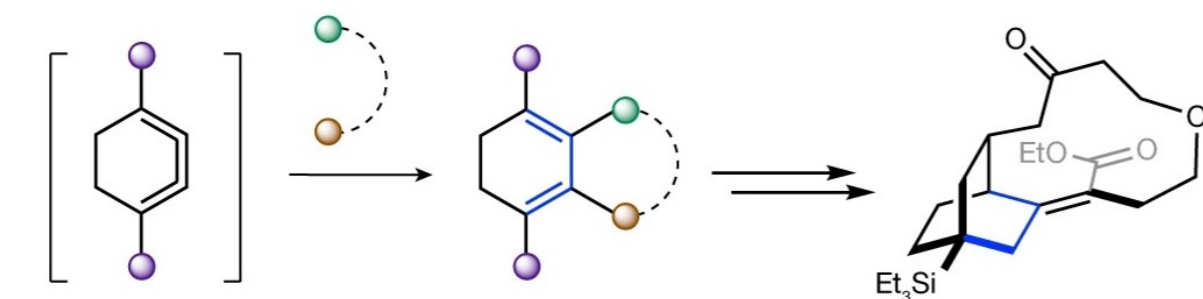


Diastereoselectivity?

Competition Experiment



Strain-promoted reactions of 1,2,3-cyclohexatriene and its derivatives



From readily accessible precursors under mild conditions

Diverse highly functionalized 1,3-dienes

Unconventional entryway to complex scaffolds

Synthetic elaboration

[4+2] Cycloadditions

[3+2] Cycloadditions

[2+2] Cycloadditions

Nu trapping

symmetrical and unsymmetrical triene

